

Types of Chemical Reactions – Complete the note taking process for the example as we study chemical reactions.

1. **Synthesis Reactions** – two or more substances combine to form a new compound. (eg.  $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$ )

Symbol Equation:  $\text{A} + \text{B} \rightarrow \text{AB}$

Example Equation:  $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{g})$

2. \_\_\_\_\_ Reaction –

Symbol Equation:

Example Equation:

3. \_\_\_\_\_ Reaction –

Symbol Equation:

Example Equation:

4. \_\_\_\_\_ Reaction –

Symbol Equation:

Example Equation:

5. \_\_\_\_\_ Reaction –

Symbol Equation:

Example Equation:

# Note Taking Chemical Reactions Answer Guide

**David Scott**



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