

Acetic Acid Indicator: Without Indicator add to _____ (mL)

1. _____ Indicator: _____ (mL) _____ (mL) _____ (mL)

2. _____ Indicator: _____ (mL) _____ (mL) _____ (mL)

3. _____ Indicator: _____ (mL) _____ (mL) _____ (mL)

Procedure: Add 10 mL "A" to each, "B" for each, and "C" for each.

1. _____ 2. _____ 3. _____

Substance	Acid (A), Base (B), or Salt (C)	Indicator with pH (or pH)	Conductivity (C or HCl)
H ₂ SO ₄	A	1	1
HNO ₃	A	1	1
HCl	A	1	1
NaOH	B	1	1
KOH	B	1	1
NaCl	C	1	1
KCl	C	1	1
Na ₂ SO ₄	C	1	1
Na ₂ CO ₃	C	1	1
Na ₂ C ₂ O ₄	C	1	1
Na ₂ SO ₃	C	1	1
Na ₂ SO ₄	C	1	1

Note Taking Guide Episode 1101 Chemistry

Thomas Griffiths



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