

Question		Expected Answers	Marks	Additional Guidance
2	a	$\text{BrO}_3^- + 5\text{Br}^- + 6\text{H}^+ \longrightarrow 3\text{Br}_2 + 3\text{H}_2\text{O}$ ✓	1	ALLOW multiples
	b	graph: Straight/diagonal line through origin OR 0,0 AND 1st order with respect to BrO_3^- ✓ initial rates data: When $[\text{Br}^-]$ is doubled, rate $\times 2$ ✓ 1st order with respect to Br^- ✓ When $[\text{H}^+] \times 2$, rate $\times 4$ (2^2) ✓ 2nd order with respect to H^+ ✓ Rate equation rate = $k [\text{BrO}_3^-] [\text{Br}^-] [\text{H}^+]^2$ ✓	1 4 1	ANNOTATIONS MUST BE USED Both explanation and 1st order required for mark DO NOT ALLOW diagonal line OR straight line OR constant gradient on its own (no mention of origin OR 0,0) ALLOW 'As BrO_3^- doubles, rate doubles' AND 1st order ALLOW rate is proportional to concentration AND 1st order Mark order and explanation independently Mark order first, then explanation ALLOW ECF from candidate's orders above

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