

(iii) What does $x\text{H}_2\text{O}$ represent in the formula $\text{Al}_2(\text{SO}_4)_3 \cdot x\text{H}_2\text{O}$?

Water of crystallisation.

[1]

H_2O crystals to constant mass.

has a mass of 5.76 g.

value of x .

+



$$\frac{5.76\text{g}}{18}$$

$$= 0.32\text{mol}$$

$$\frac{0.02}{=}$$

$$16$$

$x =$



Ocr F321 June 2013 Past Paper

G Psacharopoulos



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