

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $\text{C}_{10}\text{H}_{18}\text{O}$ (molecular)

Formula represents the number of _____ of _____
in a single molecule.

Empirical $\text{C}_5\text{H}_9\text{O}$ (molecular)

Formula represents the simplest ratio of
positive and negative ions in a formula
unit.

Atomic Mass

The average mass of atoms relative to hydrogen
atoms together.

Isotopes

^{12}C and ^{13}C isotopes

Isotopes are chemically identical
atoms.

Abundance

^{12}C is abundant isotopes.

Isotopes are chemically identical

^{12}C is abundant

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J Dewey



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