

Periodic Table Basics

Step 1: Complete the square for each element by filling in the atomic number, name, & atomic mass.

Step 2: Determine the number of protons, neutrons, and electrons in an atom of each element.

Step 3: Identify if the element is a solid, liquid, or gas at room temperature.

Step 4: Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

Step 5: List at least three physical or chemical properties for each element.

Step 6: List at least three uses for each element.

Step 7: Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

Step 8: Use the following colors to shade in the square for each element. You should **ONLY** color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na
Orange = B & Al

Pink = O & S
Red = C & Si

Blue = Be & Mg
Tan = N & P

Purple = F & Cl
Yellow = He, Ne, & Ar

Step 9: Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

① B 	P= ② N= E=	③ O S O L O G
	M.P. = B.P. = ④	
Properties ⑤		
Uses ⑥		
Bohr Diagram Lewis Structure		
 ⑦ B		

Done? Answer the questions on the worksheet using the information on your Periodic Table!

Physical Science If8767 Answers

S Ben Porath



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Senior Physical Science ,1969 Modern Physical Science William Osbert Brooks,George R. Tracy,1966 **Physical Science** National Learning Corporation,2018 The DSST Physical Science Passbook R prepares candidates for the DSST exam which enables schools to award credit for knowledge acquired outside the normal classroom environment It provides a series of informational texts as well as hundreds of questions and answers in the areas that will likely be covered on your upcoming exam including but not limited to physics electricity and magnetism Glossyr chemical reactions atomic structure and more

Physical Science Gr. 10 J. M. Lucas,D. Wright,2007* Experiences in Physical Science Laidlaw Brothers,1983 *A First Course in Physical Science* Robert P. Bauman,1987 This text provides an understanding of the foundations and structure of physical science by emphasizing science as a search for truth rather than an accumulation of facts It develops the subject through concrete examples such as inclined planes and levers speeds and displacements progressing to consideration of forces and the concept of inertia and the idea of energy Similarly a study of observable chemical reactions advances to the ways in which atoms combine separate and displace one another and how observable masses of reactants and products illustrate how the atoms are combining The fundamental ideas are applied to astronomy optics geology music and the chemical compounds of life processes The wide variety of end of chapter problems and multiple choice questions reinforce comprehension of each topic *Physical Science, a Search for Understanding. Teacher's Introduction and Answer*

Key Walter R. Brown,Norman D. Anderson,1972 **Physical Science Quiz** ,1991 **An Approach to Physical Science**

Physical Science for Nonscience Students Project,1969 **Physical Science Workbook** ,1995 **Physical Science** Robert H. Marshall,Donald H. Jacobs,Media Materials, Inc,1987 Physical World an Introduction to Physical Science Donovan Hadaway,Bob Jones University Press,2005-06 The Physical World An Introduction to Physical Science for Christian Schools was written for you the curious student We have filled this textbook with answers to puzzling questions about why things happen and how things work But this text should do more than simply answer your questions It is intended to stimulate new questions that will cause you to expand your knowledge You will be introduced to realms that you have never before explored This book will take you inside an atom let you see what happens in solids liquids and gases help you to discover the forces that make things move or keep things from moving and show you forms of matter and energy that scientists are just beginning to understand Introduction **Principles of physical Science** ,1971 **The Best of the**

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