

## 5 - Periodicity Activity

20 19 18 17 16 15 14 13 12 11 10 9 8 7 6 5 4 3 2 1

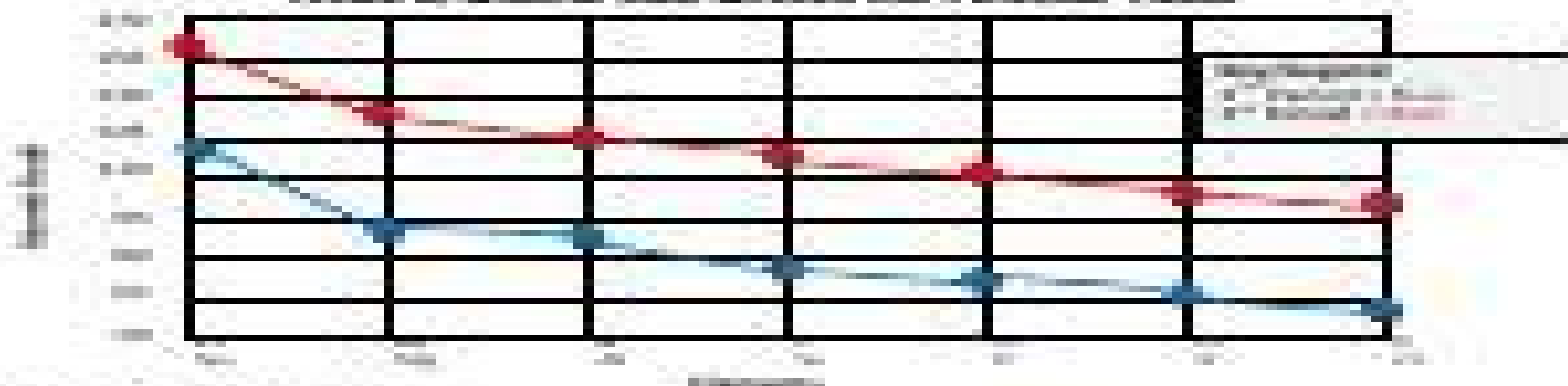
Instructions: Using the graphs above fill the boxes of atomic properties below:

1. Choose which graph best fits the property. Sketch it's data points graphically.
2. Is a slight downward trend in this graph Period? If not, is the value graph.
3. Label the box as 'not periodic' or 'periodic'.
4. Answer all questions.

	1	2	3	4	5	6	7	8
1	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
2	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
3	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
4	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
5	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
6	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
7	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
8	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
9	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
10	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity

Now label the trends above in the following way:

Trend in Atomic Size across the Periodic Table



Electronegativity trend questions

1. Following the trend in atomic size across the periodic table, decreases
2. This causes two more than just more electronegativity properties differences. They give all elements, making electronegativity trend in electronegativity trend.

Electronegativity trend questions

1. Electronegativity trend in the periodic table is decreasing (e.g. Na, Cl, Br, I, At). The trend in electronegativity across the periodic table is decreasing
2. This causes two more than just more electronegativity properties differences. They give all elements, making electronegativity trend.

# Periodicity Activity Okstate Answer Key

**L Darling-Hammond**



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