

TYBSC SEM VI PHYSICAL CHEMISTRY (MODEL MCQ's)

1	Which of the following is a uni-univalent electrolyte? a) KCl b) CuCl_2 c) AlCl_3 d) MgCl_2
2	The ionic strength of a solution of molality m_i and valency of the ion Z_i is ____ a) $\mu = \frac{1}{2} m_i Z_i^2$ b) $\mu = \sum m_i Z_i^2$ c) $\mu = \frac{1}{2} \sum Z_i^2$ d) $\mu = \frac{1}{2} \sum m_i Z_i^2$
3	Which of the following is an example of electrolyte concentration cell with transference? a) $\text{Cu} \text{CuSO}_4 \text{AgNO}_3 \text{Ag}$ b) $\text{Pt}, \text{H}_{2(g)} \text{HCl}_{(aq)} \text{AgCl}_{(s)} \text{Ag}$ c) $\text{Pt}, \text{H}_{2(g)} \text{HCl}_{(aq)}(a_1) \text{AgNO}_3(a_2) \text{Ag}$ d) $\text{Ag} - \text{AgCl} \text{NaCl}(a_1) \text{NaCl}(a_2) \text{AgCl} \cdot \text{Ag}$
4	A concentration cell which has salt bridge in its representation is ____ a) electrolyte concentration cell with transference b) Electrode concentration cell c) Electrolyte concentration cell without transference d) Gas concentration cell
5	The total emf of the electrolyte concentration cell with transference containing the liquid junction potential is given by..... a) $E_j = E_{\text{oxidation}} + E_{\text{reduction}}$ b) $E_{\text{cell}} = E_{\text{oxidation}} + E_{\text{reduction}} + E_j$ c) $E_{\text{cell}} = E_{\text{oxidation}} - E_{\text{reduction}} - E_j$ d) $E_j = E_{\text{oxidation}} - E_{\text{reduction}}$
6	The activity 'a' is related to the activity coefficient ' γ ' by the equation____ a) $a = \frac{m}{\gamma}$ b) $\gamma = a - m$ c) $a = m + \gamma$ d) $a = m\gamma$

Physical Chemistry Multiple Choice Questions

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Multiple Choice Questions in Physical Chemistry for Undergraduate Students Dr Kusum Baser, 2022-08-25 Objective questions are very much useful for practice of particular topic Multiple choice questions are important to practice any topic to solve objective questions very deep study and knowledge of specific topic is needed This book helps the students to do practice in such a way so they can do preparation for competitive exams This book covers important topics of physical chemistry for undergraduate students Physical Chemistry: A Level Multiple Choice Questions Margaret Marion Langley Oblitas, 1970-01-01

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carefully crafted to help students increase their confidence when using physics and mathematics to answer fundamental questions about the structure of molecules how chemical reactions take place and why materials behave the way they do

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AQA A-level Year 2 Chemistry Student Guide: Physical chemistry 2 Alyn G. McFarland, Nora Henry, 2016-04-04 Exam Board AQA Level A level Subject Chemistry First Teaching September 2016 First Exam June 2017 Written by experienced examiners Alyn McFarland and Nora Henry this Student Guide for Chemistry Helps students identify what they need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions Develops independent learning and research skills

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Pharmaceutical Physical Chemistry: Theory and Practices S. K. Bhasin, Physical chemistry is a compulsory paper offered to all the students of pharmacy There is a dearth of good books that exclusively cover the syllabi of physical chemistry offered to pharmacy courses *Pharmaceutical Physical Chemistry Theory and Practices* has been designed considering their requirements laid down by AICTE and other premier institutes universities Apart from the theory 20 most common laboratory experiments have been included to make this book a unique offering to the students of pharmacy

Physical Chemistry of Polymers Sebastian Seiffert, 2023-01-30 This book introduces the concepts of physical chemistry of polymers in a format targeted for a blended learning approach It provides a basis to bridge polymer chemistry which targets microscopic chain structures and polymer engineering which targets macroscopic material properties and functions Topics covered are single chain statistics multi chain interactions and chain dynamics both from a viewpoint of structure properties mostly mechanical ones and their interrelation In all that the author encourages the reader to think conceptually Explains complex facts through simplifying models diagrams and illustrations Accessible to chemists chemical engineers materials scientists and physicists Tailored content for an interactive blended learning format

The Pearson Guide To Physical Chemistry For The Aipmt Singhal Atul, 2011-09

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