

## 2.6 - Periodic Trends worksheet

Name \_\_\_\_\_

### I. Short answer

1. Label each group on the periodic table.
2. Explain what valence electrons are.
3. Explain what makes an element stable?

### II. Answer the questions using the symbols on the periodic table

\_\_\_\_\_ 1. What elements are in period 5?

What element(s) is a

\_\_\_\_\_ 2. halogens

\_\_\_\_\_ 6. transition metals

\_\_\_\_\_ 3. lanthanides

\_\_\_\_\_ 7. alkali metals

\_\_\_\_\_ 4. alkaline earth metals

\_\_\_\_\_ 8. actinides

\_\_\_\_\_ 5. noble gases

What elements are in the sublevel

\_\_\_\_\_ 9. s

\_\_\_\_\_ 11. d

\_\_\_\_\_ 10. p

\_\_\_\_\_ 12. f

What elements are

\_\_\_\_\_ 13. metals,

\_\_\_\_\_ 14. nonmetals

\_\_\_\_\_ 15. metalloids,

\_\_\_\_\_ 16. Which element is the most reactive nonmetal?

\_\_\_\_\_ 17. Which element is the most reactive metal?

\_\_\_\_\_ 18. Which element below is the least reactive?

\_\_\_\_\_ 19. Which element has the largest radius in period 4?

\_\_\_\_\_ 20. Which element has the largest radius in the Alkali metals family?

\_\_\_\_\_ 21. Which element has the highest ionization energy in period four?

\_\_\_\_\_ 22. Which elements has the second largest atomic number?

\_\_\_\_\_ 23. Which element has the largest atomic mass?

\_\_\_\_\_ 24. Which element is located at period 4, group 15?

\_\_\_\_\_ 25. Which element is located at period 5, group 2?

\_\_\_\_\_ 27. Which elements will form +1 ions?

\_\_\_\_\_ 28. Which group of metals will have the same properties

\_\_\_\_\_ 29. Which group of nonmetals will have the same properties

\_\_\_\_\_ 30. On the back, define all chapter 5, section 3 vocabulary terms.

b																	r		
																	h		o
	a																		u
i																			
				e															z

		w												
					y									



# Periodic Table Chemistry Answers

**Clifford C. Houk, Richard Post**



## Periodic Table Chemistry Answers:

Chemistry Questions - Answers for Intermediate Mr. Ajeet Singh, 2018-05-13 A question answer formatted book designed for intermediate level chemistry students covering theory based and numerical questions to aid learning and exam preparation

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### **Table of Contents Periodic Table Chemistry Answers**

1. Understanding the eBook Periodic Table Chemistry Answers
  - The Rise of Digital Reading Periodic Table Chemistry Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Chemistry Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Chemistry Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Chemistry Answers
  - Personalized Recommendations

- Periodic Table Chemistry Answers User Reviews and Ratings
- Periodic Table Chemistry Answers and Bestseller Lists
- 5. Accessing Periodic Table Chemistry Answers Free and Paid eBooks
  - Periodic Table Chemistry Answers Public Domain eBooks
  - Periodic Table Chemistry Answers eBook Subscription Services
  - Periodic Table Chemistry Answers Budget-Friendly Options
- 6. Navigating Periodic Table Chemistry Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Periodic Table Chemistry Answers Compatibility with Devices
  - Periodic Table Chemistry Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Table Chemistry Answers
  - Highlighting and Note-Taking Periodic Table Chemistry Answers
  - Interactive Elements Periodic Table Chemistry Answers
- 8. Staying Engaged with Periodic Table Chemistry Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Table Chemistry Answers
- 9. Balancing eBooks and Physical Books Periodic Table Chemistry Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table Chemistry Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Chemistry Answers
  - Setting Reading Goals Periodic Table Chemistry Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Chemistry Answers
  - Fact-Checking eBook Content of Periodic Table Chemistry Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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