

CHAPTER 5: PERIODIC

The Periodic Law

SECOND PART

Student's Assistant Answered the following questions for the second part of the lab.

1. a) Consider the halogen atoms with 10 protons, and the elements adjacent to the halogen group.

- _____ a. The first is the atomic number?
- _____ b. The last is the atomic number?
- _____ c. Is the element in period 10 or another period, rather than period 10? Is the element number 10 the atomic number?

2. Consider the elements with electron configurations for $1s^2 2s^2 2p^2$.

- _____ a. The first element is the atomic number?
- _____ b. It is in a halogen, what is the atomic number?
- _____ c. The group number will equal $10 + 2 = 12$. Is it?

3. The halogen atoms are chemically similar, e.g., $1s^2 2s^2 2p^2$.

- _____ a. The halogen atoms are chemically similar, e.g., $1s^2 2s^2 2p^2$.
- _____ b. The halogen atoms are other elements, chemical formulae.
- _____ c. The first is the atomic number.
- _____ d. The last is the atomic number.
- _____ e. The first is the atomic number.
- _____ f. The last is the atomic number.

4. The first element is the halogen, rather than the first of the group 10. The first is the atomic number.

- _____ a. The first is the atomic number.
- _____ b. The last is the atomic number.
- _____ c. The first is the atomic number.
- _____ d. The first is the atomic number.
- _____ e. The first is the atomic number.
- _____ f. The first is the atomic number.
- _____ g. The first is the atomic number.
- _____ h. The first is the atomic number.
- _____ i. The first is the atomic number.
- _____ j. The first is the atomic number.
- _____ k. The first is the atomic number.
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- _____ t. The first is the atomic number.
- _____ u. The first is the atomic number.
- _____ v. The first is the atomic number.
- _____ w. The first is the atomic number.
- _____ x. The first is the atomic number.
- _____ y. The first is the atomic number.
- _____ z. The first is the atomic number.

Periodic Law Mixed Review Answer Key

Mark Steyvers



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