

# Percent Yield

1. In a certain chemical reaction, 4.00 g of reactant A (molar mass 40.0 g/mol) reacts with 1.00 g of reactant B (molar mass 20.0 g/mol) to produce 2.00 g of product C (molar mass 60.0 g/mol). What is the percent yield of product C?
2. A chemist reacts 10.0 g of reactant X (molar mass 100.0 g/mol) with 5.0 g of reactant Y (molar mass 50.0 g/mol) to produce 3.0 g of product Z (molar mass 150.0 g/mol). What is the percent yield of product Z?
3. A reaction produces 12.0 g of product D from 6.0 g of reactant E (molar mass 120.0 g/mol) and 3.0 g of reactant F (molar mass 60.0 g/mol). What is the percent yield of product D?
4. A chemist reacts 8.0 g of reactant G (molar mass 80.0 g/mol) with 4.0 g of reactant H (molar mass 40.0 g/mol) to produce 2.0 g of product I (molar mass 120.0 g/mol). What is the percent yield of product I?
5. A reaction produces 15.0 g of product J from 7.5 g of reactant K (molar mass 150.0 g/mol) and 3.75 g of reactant L (molar mass 75.0 g/mol). What is the percent yield of product J?
6. A chemist reacts 12.0 g of reactant M (molar mass 120.0 g/mol) with 6.0 g of reactant N (molar mass 60.0 g/mol) to produce 4.0 g of product O (molar mass 180.0 g/mol). What is the percent yield of product O?
7. A reaction produces 18.0 g of product P from 9.0 g of reactant Q (molar mass 180.0 g/mol) and 4.5 g of reactant R (molar mass 90.0 g/mol). What is the percent yield of product P?
8. A chemist reacts 16.0 g of reactant S (molar mass 160.0 g/mol) with 8.0 g of reactant T (molar mass 80.0 g/mol) to produce 5.0 g of product U (molar mass 240.0 g/mol). What is the percent yield of product U?
9. A reaction produces 20.0 g of product V from 10.0 g of reactant W (molar mass 200.0 g/mol) and 5.0 g of reactant X (molar mass 100.0 g/mol). What is the percent yield of product V?
10. A chemist reacts 24.0 g of reactant Y (molar mass 240.0 g/mol) with 12.0 g of reactant Z (molar mass 120.0 g/mol) to produce 6.0 g of product AA (molar mass 360.0 g/mol). What is the percent yield of product AA?

# Percent Yield Skill Practice 44

**Mark Steyvers**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

**Percent Yield Skill Practice 44:**

## **Percent Yield Skill Practice 44** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Percent Yield Skill Practice 44**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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