

## Polyatomic Ions WS

Ammonium ( $\text{NH}_4$ )<sup>1+</sup>  
 Carbonate ( $\text{CO}_3$ )<sup>2-</sup>  
 Chromate ( $\text{CrO}_4$ )<sup>2-</sup>

Dichromate ( $\text{Cr}_2\text{O}_7$ )<sup>2-</sup>  
 Hydroxide ( $\text{OH}$ )<sup>1-</sup>  
 Nitrate ( $\text{NO}_3$ )<sup>1-</sup>

Nitrite ( $\text{NO}_2$ )<sup>1-</sup>  
 Sulfate ( $\text{SO}_4$ )<sup>2-</sup>  
 Sulfite ( $\text{SO}_3$ )<sup>2-</sup>

Phosphate ( $\text{PO}_4$ )<sup>3-</sup>

Write the NAME of each of the following compounds.

1.  $(\text{NH}_4)\text{Cl}$  Ammonium Chloride
2.  $\text{Be}(\text{SO}_4)$  Beryllium Sulfate
3.  $(\text{NH}_4)_3\text{N}$  Ammonium Nitride
4.  $\text{MgCl}_2$  Magnesium Chloride
5.  $\text{NH}_4(\text{NO}_3)$  Ammonium Nitrate
6.  $\text{Sr}_3(\text{PO}_4)_2$  Strontium Phosphate
7.  $\text{Zn}(\text{CrO}_4)_2$  Zinc Dichromate
8.  $\text{K}_2(\text{Cr}_2\text{O}_7)$  Potassium Dichromate

9.  $\text{Ga}(\text{ClO}_3)_3$  Gallium Chlorate
10.  $\text{Cu}(\text{OH})$  Copper I Hydroxide
11.  $(\text{NH}_4)_3(\text{PO}_4)$  Ammonium Phosphate
12.  $\text{Fe}(\text{SO}_4)$  Iron II Sulfate
13.  $\text{Mg}(\text{NO}_2)_2$  Magnesium Nitrite
14.  $(\text{NH}_4)\text{NO}_2$  Ammonium Nitrite
15.  $\text{Na}_2(\text{Cr}_2\text{O}_7)$  Sodium Dichromate
16.  $\text{Na}(\text{OH})$  Sodium Hydroxide

Write the CHEMICAL FORMULA for each of the given NAMES.....cross charges

17. calcium carbonate  $\text{Ca}^{2+} (\text{CO}_3)^{2-}$  =  $\text{Ca}_2(\text{CO}_3)_2$  =  $\text{CaCO}_3$
18. barium nitrate  $\text{Ba}^{+2} \text{NO}_3^{-1}$  =  $\text{Ba}(\text{NO}_3)$
19. ammonium sulfate  $\text{NH}_4^{+1} \text{SO}_4^{-2}$  =  $(\text{NH}_4)_2 \text{SO}_4$
20. aluminum hydroxide  $\text{Al}^{+3} \text{OH}^{-1}$  =  $\text{Al}(\text{OH})_3$
21. calcium phosphate  $\text{Ca}^{+2} \text{PO}_4^{-3}$  =  $\text{Ca}_3(\text{PO}_4)_2$
22. cesium nitrate  $\text{Cs} \text{NO}_2^{-1}$  =  $\text{CsNO}_3$
23. sodium nitrite  $\text{Na}^{+1} \text{NO}_3^{-1}$  =  $\text{NaNO}_3$
24. calcium sulfate  $\text{Ca}^{+2} \text{SO}_4^{-2}$  =  $\text{CaSO}_4$
25. beryllium sulfate  $\text{Be} \text{SO}_4^{-2}$  =  $\text{BeSO}_4$
26. sodium carbonate  $\text{Na}^{+1} \text{CO}_3^{-2}$  =  $\text{Na}_2\text{CO}_3$
27. magnesium phosphate  $\text{Mg}^{+2} \text{PO}_4^{-3}$  =  $\text{Mg}_3(\text{PO}_4)_2$
28. calcium phosphate  $\text{Ca}^{+2} \text{PO}_4^{-3}$  =  $\text{Ca}_3(\text{PO}_4)_2$

# Polyatomic Ions Ws

**Rosina Ehmann**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

## **Polyatomic Ions Ws:**

## Reviewing **Polyatomic Ions Ws**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Polyatomic Ions Ws**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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