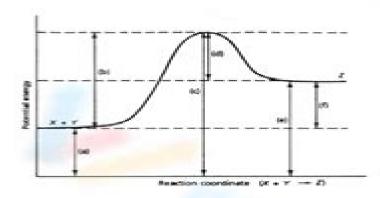
Potential Energy Diagram Worksheet ANSWERS



- Which of the letters a-f in the diagram represents the potential energy of the products?
- Which letter indicates the potential energy of the activated complex?
- Which letter indicates the potential energy of the reactants?
- Which letter indicates the activation energy?
- Which letter indicates the heat of reaction?
- Is the reaction exothermic or endothermic? endo
- Which letter indicates the activation energy of the reverse reaction?
- Which letter indicates the heat of reaction of the reverse reaction?
- Is the reverse reaction exothermic or endothermic? __exo__



- The PE of the reactants of the forward reaction is about 80 kilojoules.
- The PE of the products of the forward reaction is about __160__ kilojoules.
- The PE of the activated complex of the forward reaction is about <u>240</u> kilojoules.
- The activation energy of the forward reaction is about __160_ kilojoules.
- The heat of reaction (AH) of the forward reaction is about __+80_ kilojoules.
- 6. The forward reaction is endothermic

(endothermic or exothermic).

- The PE of the reactants of the reverse reaction is about 160 kilojoules.
- The PE of the products of the reverse reaction is about 80 kilojoules.
- The PE of the activated complex of the reverse reaction is about <u>240</u> kilojoules.
- The activation energy of the reverse reaction is about __80_ kilojoules.
- 11. The heat of reaction (AH) of the reverse reaction is about -80 kilojoules.
- The reverse reaction is ____exothermic_____ (endothermic or exothermic).

Potential Energy Diagram Answer Key

Jicheng Xie

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