

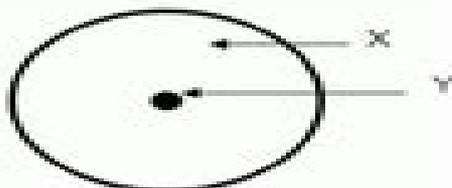
## Test Periodic Table and Trends SCH3U

Name: \_\_\_\_\_

Multiple Choice: Select the most correct answer (20K)

1. The mass number of an atom is determined by:

- a. the number of protons only
- b. adding the neutrons and protons
- c. adding the neutrons and electrons
- d. adding the protons and electrons



2. The letter "X" in the diagram above marks the:

- a. Region of greatest density
- b. Location of the protons
- c. Location of the neutrons
- d. Location of the electrons
- e. Nucleus

3. Most of the mass of the atom can be found in the:

- a. charges
- b. electron cloud
- c. electrons
- d. nucleus

4. Of the basic atomic particles, the one that would be attracted to a negatively charged metallic plate is the:

- a. proton
- b. neutron
- c. electron
- d. none of the above

5. Isotopes (such as hydrogen-1, hydrogen-2, and hydrogen-3) are atoms of the same element that differ in:

- a. the number of protons in the nucleus
- b. the number of electrons
- c. the atomic number
- d. the number of neutrons in the nucleus
- e. the mass number

6. Which of the following accurately summarizes the isotope Argon-40?

- a. 40 protons, 40 electrons, 18 neutrons
- b. 18 protons, 18 electrons, 22 neutrons
- c. 39.95 protons, 39.95 electrons, 21.05 neutrons
- d. 22 protons, 22 electrons, 18 neutrons

# Periodic Trends Test

**Neil D. Jespersen, Pamela Kerrigan**



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