

**Periodic Trends**

1. Explain the following trends:

- Atomic Radius
- Electronegativity
- Ionization Energy

2. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

3. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

4. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

5. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

6. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

7. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

8. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

9. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

10. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Period: \_\_\_\_\_

### Periodic Trends Test Study Guide

**Section 1: Atomic Radius and Electronegativity**

1. Explain the following trends:

- Atomic Radius: Increases from top-left to bottom-right.
- Electronegativity: Increases from bottom-left to top-right.

2. Explain the following trends:

- Atomic Radius: Increases from top-left to bottom-right.
- Electronegativity: Increases from bottom-left to top-right.

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10. Explain the following trends:

- Atomic Radius: Increases from top-left to bottom-right.
- Electronegativity: Increases from bottom-left to top-right.

# Periodic Table Study Guide Word Search

**N Colangelo**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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