

Periodic Table of the Elements

1 IA H Hydrogen (1.008)																	18 VIIIA He Helium (4.003)		
3 IIA Li Lithium (6.941)	4 IIA Be Beryllium (9.012)													5 IIIA B Boron (10.81)	6 IIIA C Carbon (12.011)	7 IVA N Nitrogen (14.007)	8 IVA O Oxygen (15.999)	9 VA F Fluorine (18.998)	10 VA Ne Neon (20.180)
11 IA Na Sodium (22.990)	12 IIA Mg Magnesium (24.305)													13 IIIA Al Aluminum (26.982)	14 IIIA Si Silicon (28.086)	15 IVA P Phosphorus (30.974)	16 IVA S Sulfur (32.06)	17 VA Cl Chlorine (35.45)	18 VA Ar Argon (39.948)
19 IA K Potassium (39.098)	20 IIA Ca Calcium (40.078)	21 IIB Sc Scandium (44.956)	22 IIB Ti Titanium (47.88)	23 IIB V Vanadium (50.942)	24 IIB Cr Chromium (51.996)	25 IIB Mn Manganese (54.938)	26 IIB Fe Iron (55.845)	27 IIB Co Cobalt (58.933)	28 IIB Ni Nickel (58.693)	29 IB Cu Copper (63.546)	30 IB Zn Zinc (65.38)	31 IIB Ga Gallium (69.723)	32 IIB Ge Germanium (72.63)	33 IIIA As Arsenic (74.922)	34 IIIA Se Selenium (78.96)	35 IIIA Br Bromine (79.904)	36 IIIA Kr Krypton (83.798)		
37 IA Rb Rubidium (85.468)	38 IIA Sr Strontium (87.62)	39 IIB Y Yttrium (88.906)	40 IIB Zr Zirconium (91.224)	41 IIB Nb Niobium (92.906)	42 IIB Mo Molybdenum (95.94)	43 IIB Tc Technetium (98.906)	44 IIB Ru Ruthenium (101.07)	45 IIB Rh Rhodium (102.91)	46 IIB Pd Palladium (106.37)	47 IB Ag Silver (107.87)	48 IB Cd Cadmium (112.41)	49 IIB In Indium (114.82)	50 IIB Sn Tin (118.71)	51 IIIA Sb Antimony (121.76)	52 IIIA Te Tellurium (127.6)	53 IIIA I Iodine (126.91)	54 IIIA Xe Xenon (131.29)		
55 IA Cs Cesium (132.91)	56 IIA Ba Barium (137.33)	57-71 Lanthanides	72 IIB Hf Hafnium (178.49)	73 IIB Ta Tantalum (180.95)	74 IIB W Tungsten (183.84)	75 IIB Re Rhenium (186.21)	76 IIB Os Osmium (190.23)	77 IIB Ir Iridium (192.22)	78 IIB Pt Platinum (195.08)	79 IB Au Gold (196.97)	80 IB Hg Mercury (200.59)	81 IIB Tl Thallium (204.38)	82 IIB Pb Lead (207.2)	83 IIIA Bi Bismuth (208.98)	84 IIIA Po Polonium (209)	85 IIIA At Astatine (210)	86 IIIA Rn Radon (222)		
87 IA Fr Francium (223)	88 IIA Ra Radium (226)	89-103 Actinides	104 IIB Rf Rutherfordium (261)	105 IIB Db Dubnium (262)	106 IIB Sg Seaborgium (266)	107 IIB Bh Bohrium (264)	108 IIB Hs Hassium (277)	109 IIB Mt Meitnerium (268)	110 IIB Ds Darmstadtium (271)	111 IB Rg Roentgenium (272)	112 IB Cn Copernicium (285)	113 IIB Nh Nihonium (284)	114 IIB Fl Flerovium (289)	115 IIIA Mc Moscovium (288)	116 IIIA Lv Livermorium (293)	117 IIIA Ts Tennessine (289)	118 IIIA Og Oganesson (294)		

57 La Lanthanum (138.91)	58 Ce Cerium (140.12)	59 Pr Praseodymium (140.91)	60 Nd Neodymium (144.24)	61 Pm Promethium (145)	62 Sm Samarium (150.36)	63 Eu Europium (151.96)	64 Gd Gadolinium (157.25)	65 Tb Terbium (158.93)	66 Dy Dysprosium (162.50)	67 Ho Holmium (164.93)	68 Er Erbium (167.26)	69 Tm Thulium (168.93)	70 Yb Ytterbium (173.05)	71 Lu Lutetium (174.97)
89 Ac Actinium (227)	90 Th Thorium (232.04)	91 Pa Protactinium (231.04)	92 U Uranium (238.03)	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (262)

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A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the 'Disha Experts' text.

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Structuralist Knowledge Representation , 2022-05-16

Organic Chemistry Michael B. Smith, 2011-06-29 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts Organic Chemistry An Acid Base Approach provides a framework for understanding the subject that goes beyond mere memorization The individual steps in many important mechanisms rely on acid base reactions and the ability to see these relationships makes understanding organic chemistry easier Using several techniques to develop a relational understanding this textbook helps

students fully grasp the essential concepts at the root of organic chemistry Providing a practical learning experience with numerous opportunities for self testing the book contains Checklists of what students need to know before they begin to study a topic Checklists of concepts to be fully understood before moving to the next subject area Homework problems directly tied to each concept at the end of each chapter Embedded problems with answers throughout the material Experimental details and mechanisms for key reactions The reactions and mechanisms contained in the book describe the most fundamental concepts that are used in industry biological chemistry and biochemistry molecular biology and pharmacy The concepts presented constitute the fundamental basis of life processes making them critical to the study of medicine Reflecting this emphasis most chapters end with a brief section that describes biological applications for each concept This text provides students with the skills to proceed to the next level of study offering a fundamental understanding of acids and bases applied to organic transformations and organic molecules

Physical Science in the Modern World Jerry Marion, 2012-12-02 Physical Science in the Modern World surveys the whole range of the non biological sciences This book explores the significant ideas and concepts in chemistry physics astronomy geology and meteorology with emphasis on how these sciences bear strongly upon one another and how the basic principles are applied to each Organized into three part encompassing 29 chapters this book starts with an overview of the fundamental building blocks of matter and explains how they are assembled to form molecules rocks minerals and the Earth This text then examines the basic concepts of physical science by exploring the fundamental principles that govern all physical processes and we see how they relate to various everyday occurrences Other chapters consider how modern chemistry affects the world we live in and explain how the development of semiconductor materials has led in the development of miniature electronics This book is a valuable resource for physicists chemists astronomers geologists and meteorologists

A New Approach to ICSE Chemistry for Class X (A.Y. 2023-24) Onward V. K. Sally, 2023-05-20 A New Approach to ICSE Chemistry for Class X has been revised in accordance with the latest syllabus prescribed by the Council for the Indian School Certificate Examination New Delhi for Class X The main strength of the book lies in the presentation of scientific content which has been so arranged that the topics are linked with each other and do not cause any unnecessary burden on the mind of the student Salient features of this book are as follows Clear simple and easy language A large number of chemical reactions along with experimental observations Full sized diagrams as expected from a student in the examination Topic wise Video Lectures as a support for effective learning Concepts supplemented by suitable day to day examples Periodic Table showing mass number atomic number of various elements along with the list of actual names of the elements Highlighted important terms and definitions Well designed exercises to assess conceptual reasoning skill and application based learning as per the latest recommendations of ICSE board Solved numerical problems at the end of each chapter to help the students solve numericals on their own A chapter on Practical Chemistry to help students in their laboratory work The latest Solved ICSE Specimen Paper has been

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