

TEACHER INSTRUCTIONS

These colorful reading articles are to get students engaged in reading science!

These reading articles are so colorful and such an important step in the learning process!

Some suggestions for use:

- "Chemistry" reading and notes are done at home with more **ASSIGNMENT** work
- Read information before starting a new unit or topic of study
- Use during the "lecture" portion of the PE model
- Print only for side-up, one different element or used in the "Spunky Science" reading lesson information
- Use with note-taking strategies in an RCU or Science classroom
- Split up pages and read with a connecting table
- Split up pages and have the students group up to create info
- Used as test work either digitally or printed

Printing Options:

- PDF and Color Version are available depending on your preferences
- Print on PDF to be placed into a journal
- Print as a "Double" form for 10 min journals or three or four pages
- Print as full pages and laminate for use as posters or in small groups

I hope you like these! I will be reading...



PERIODIC TABLE DUMP

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← Increasing electronegativity
increasing ionization energy
increasing electron affinity

← Increasing electronegativity
increasing ionization energy
increasing electron affinity

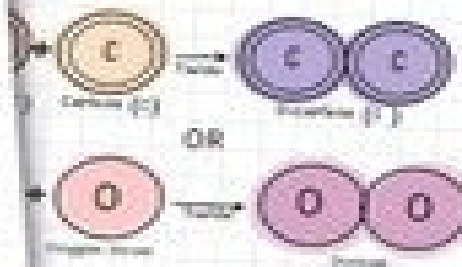
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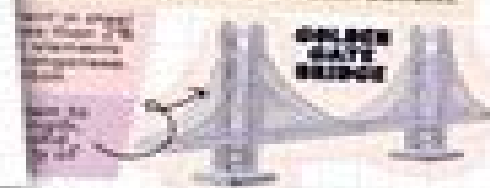
Get lost in the periodic table! Represents a specific element's location on the PT way carefully assigned based on each element's properties.

DIATOMIC MOLECULES

Many 2 or more atoms of the same element chemically combine. They are most prevalent as an element because they are the most stable of the same type of atom. They are just molecules. These types of molecules are called diatomic molecules. They are usually represented in 2 ways.



Understanding the structure of atoms helps explain how much matter will flow the way to their ability to react with other elements.



Periodic Table Note Taking Guide

Patrick Sebranek, Verne Meyer



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Table of Contents Periodic Table Note Taking Guide

1. Understanding the eBook Periodic Table Note Taking Guide
 - The Rise of Digital Reading Periodic Table Note Taking Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Note Taking Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Note Taking Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Note Taking Guide
 - Personalized Recommendations
 - Periodic Table Note Taking Guide User Reviews and Ratings
 - Periodic Table Note Taking Guide and Bestseller Lists

5. Accessing Periodic Table Note Taking Guide Free and Paid eBooks
 - Periodic Table Note Taking Guide Public Domain eBooks
 - Periodic Table Note Taking Guide eBook Subscription Services
 - Periodic Table Note Taking Guide Budget-Friendly Options
6. Navigating Periodic Table Note Taking Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Note Taking Guide Compatibility with Devices
 - Periodic Table Note Taking Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Note Taking Guide
 - Highlighting and Note-Taking Periodic Table Note Taking Guide
 - Interactive Elements Periodic Table Note Taking Guide
8. Staying Engaged with Periodic Table Note Taking Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Note Taking Guide
9. Balancing eBooks and Physical Books Periodic Table Note Taking Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Note Taking Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Note Taking Guide
 - Setting Reading Goals Periodic Table Note Taking Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Note Taking Guide
 - Fact-Checking eBook Content of Periodic Table Note Taking Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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