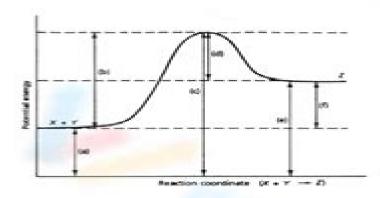
#### Potential Energy Diagram Worksheet ANSWERS



- Which of the letters a-f in the diagram represents the potential energy of the products?
- Which letter indicates the potential energy of the activated complex?
- Which letter indicates the potential energy of the reactants?
- Which letter indicates the activation energy?
- Which letter indicates the heat of reaction?
- Is the reaction exothermic or endothermic? endo
- Which letter indicates the activation energy of the reverse reaction?
- Which letter indicates the heat of reaction of the reverse reaction?
- Is the reverse reaction exothermic or endothermic? \_\_exo\_\_



- The PE of the reactants of the forward reaction is about 80 kilojoules.
- The PE of the products of the forward reaction is about \_\_160\_\_ kilojoules.
- The PE of the activated complex of the forward reaction is about <u>240</u> kilojoules.
- The activation energy of the forward reaction is about \_\_160\_ kilojoules.
- The heat of reaction (AH) of the forward reaction is about \_\_+80\_ kilojoules.
- 6. The forward reaction is endothermic

(endothermic or exothermic).

- The PE of the reactants of the reverse reaction is about 160 kilojoules.
- The PE of the products of the reverse reaction is about 80 kilojoules.
- The PE of the activated complex of the reverse reaction is about <u>240</u> kilojoules.
- The activation energy of the reverse reaction is about \_\_80\_\_ kilojoules.
- 11. The heat of reaction (AH) of the reverse reaction is about -80 kilojoules.
- The reverse reaction is \_\_\_\_exothermic\_\_\_\_\_ (endothermic or exothermic).

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Scott C. Dulebohn

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Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03

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