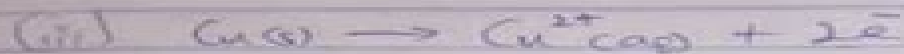


QUESTION 3 (Contd.)

(i) The set-up is used for extraction.

- (ii) • CuSO_4
• CuCl_2



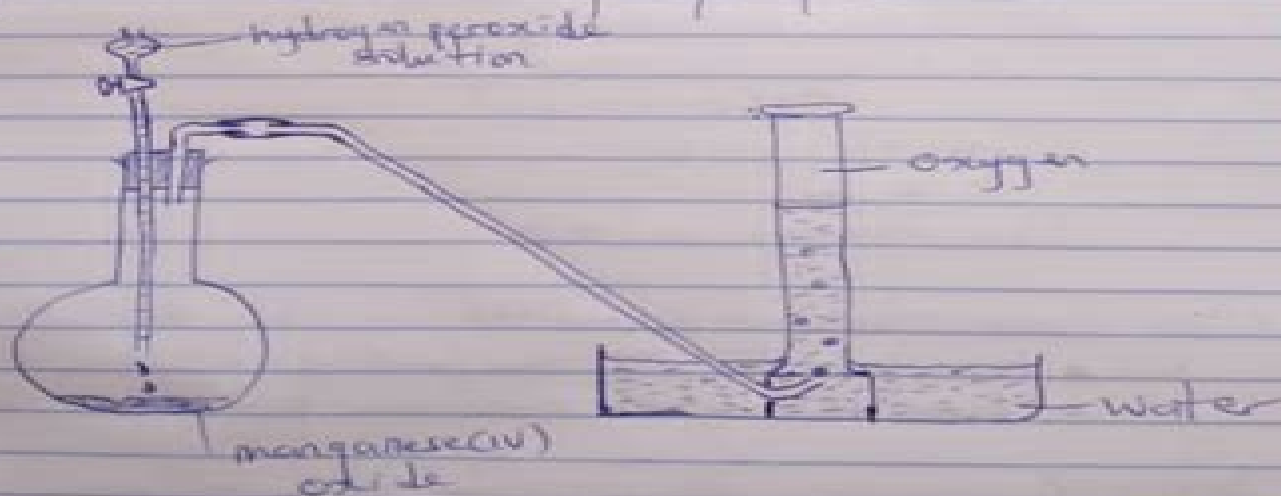
(iv) Given: $m = 3.2 \text{ g}$

$$t = 3 \text{ mins } 13 \text{ seconds} = (3 \times 60) + 13 \\ = 193 \text{ secs}$$

$$I = 50 \text{ A}$$

$$\text{Using } Z = \frac{m}{It} = \frac{3.2}{50 \times 193} = 0.00033 \text{ g/C}$$

(d) Oxygen is usually prepared in the laboratory by the oxidation of hydrogen peroxide.



The manganese(IV) oxide acts as a catalyst as the hydrogen peroxide readily decomposes on its own to liberate oxygen.

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