

More Atomic Structure Practice: Use the periodic table to help you find the answers!

Draw atomic structures and electron dot diagrams for the following atoms: H, He, Li, Be, B, C, O, F, Ne, Na, Al, Si, P, Cl, Ar

Here are the atomic structures: NOTE! Only the electrons in their shells are shown below! The number of protons equals the TOTAL number of electrons. The number of neutrons is found below.

1st Shell	H1 He2							
2nd Shell	Li3	Be4	B5	C6	N7	O8	F9	Ne10
3rd Shell	Na11	Mg12	Al13	Si14	P15	S16	Cl17	Ar18

Here are the electron dot diagrams:

I	II		III	IV	V	VI	VII	0
H •								He ••
Li •	• Be •		• B •	• C •	• N •	• O •	• F •	• Ne ••
Na •	• Mg •		• Al •	• Si •	• P •	• S •	• Cl •	• Ar ••
K •	• Ca •		• Ga •	• Ge •	• As •	• Se •	• Br •	• Kr ••
Rb •	• Sr •		• In •	• Sn •	• Sb •	• Te •	• I •	• Xe ••
Cs •	• Ba •		• Tl •	• Pb •	• Bi •	• Po •	• At •	• Rn ••

Metal
 Metalloid
 Nonmetal

How many neutrons are in the following atoms (show your work!): H, He, Li, Be, B, C, O, F, Ne, Na, Al, Si, P, Cl, Ar

Atomic mass (rounded) — Atomic Number = # neutrons

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