

**Table 2 – Data Collected for Nitric Acid Synthesis**

	1	2	3	4	5	6	7	8	9	10	11	12
1	10											
2	10											
3	10	10										
4	10	10	10	10	10	10	10	10	10	10	10	10
5	10	10	10	10	10	10	10	10	10	10	10	10
6	10	10	10	10	10	10	10	10	10	10	10	10
7	10	10	10	10	10	10	10	10	10	10	10	10
8	10	10	10	10	10	10	10	10	10	10	10	10
9	10	10	10	10	10	10	10	10	10	10	10	10
10	10	10	10	10	10	10	10	10	10	10	10	10
11	10	10	10	10	10	10	10	10	10	10	10	10
12	10	10	10	10	10	10	10	10	10	10	10	10

(a) Describe how you find the selected data represent the overall trend observed for the reaction that is studied in

(b) Compare the trend found in Table 2.

- The general trend observed from the reaction is that the reaction is exothermic.
- The general trend observed from the reaction is that the reaction is endothermic.
- The general trend observed from the reaction is that the reaction is exothermic.
- The general trend observed from the reaction is that the reaction is endothermic.

# Periodicity Of Elements Pogil Answer Key

**JG Myers**



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