

PERIODIC TABLE (ADVANCED)

Molecular Structure of Hydrogen



Basic Information

Name: Hydrogen

Symbol: H

Atomic Number: 1

Atomic Mass: 1.00794 amu

Melting Point: -259.14°C ($14.009185^{\circ}\text{K}$, $-434.45203^{\circ}\text{F}$)

Boiling Point: -252.87°C ($20.280005^{\circ}\text{K}$, -423.166°F)

Number of Protons/Electrons: 1

Number of Neutrons: 0

Classification: Non-metal

Crystal Structure: Hexagonal

Density @ 293°K : 0.08989 g/cm^3

Color: colorless

Atomic Structure

Number of Energy Levels: 1

First Energy Level: 1

Isotopes

Isotope	Half-life
H-1	Stable
H-2	Stable
H-3	12.3 years

Facts

Name Origin: from words *hydro* (water) and *genes* (generated)

Uses: Balloons, metal refining

Obtained From: mines, oil, gas wells

Molecular Structure of Carbon



Name: Carbon

Symbol: C

Atomic Number: 6

Atomic Mass: 12.0107 amu

Melting Point: 3500.0°C (3773.15°K , 6332.0°F)

Boiling Point: 4827.0°C (5100.15°K , 8700.6°F)

Number of Protons/Electrons: 6

Number of Neutrons: 6

Classification: Non-metal

Crystal Structure: Hexagonal

Density @ 293°K : 2.62 g/cm^3

Color: May be black

Atomic Structure

Number of Energy Levels: 2

First Energy Level: 2

Second Energy Level: 4

Isotopes

Isotope	Half-life
C-11	20.3 minutes
C-12	Stable
C-13	Stable
C-14	5730.0 years
C-15	2.5 seconds

Facts

Uses: steel, fibers

Obtained From: burning with insufficient oxygen

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R Barnett



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