

Periodic Table of the Elements

	+1	+2											+3	+4	+5	+6	+7	0
1	H 1.008	He 4.003											B 10.81	C 12.01	N 14.01	O 16.00	F 18.99	Ne 20.18
2	Li 6.941	Be 9.012											Al 26.98	Si 28.09	P 30.97	S 32.07	Cl 35.45	Ar 39.95
3	Na 22.99	Mg 24.31	Sc 44.96	Ti 47.88	V 50.94	Cr 52.00	Mn 54.94	Fe 55.85	Co 58.93	Ni 58.71	Cu 63.55	Zn 65.38	Ga 69.72	Ge 72.64	As 74.92	Se 78.96	Br 79.90	Kr 83.80
4	K 39.10	Ca 40.08	Sc 44.96	Ti 47.88	V 50.94	Cr 52.00	Mn 54.94	Fe 55.85	Co 58.93	Ni 58.71	Cu 63.55	Zn 65.38	Ga 69.72	Ge 72.64	As 74.92	Se 78.96	Br 79.90	Kr 83.80
5	Rb 85.47	Sr 87.62	Y 88.91	Zr 91.22	Nb 92.91	Mo 95.94	Tc 98.91	Ru 101.07	Rh 102.91	Pd 106.42	Ag 107.87	Cd 112.41	In 114.82	Sn 118.71	Sb 121.76	Te 127.60	I 126.91	Xe 131.29
6	Cs 132.91	Ba 137.33	La 138.91	Ce 140.12	Pr 140.91	Nd 144.24	Pm 144.91	Sm 150.36	Eu 151.96	Gd 157.25	Tb 158.93	Dy 162.50	Ho 164.93	Er 167.26	Tm 168.93	Yb 173.05	Lu 174.97	
7	Fr 223.02	Ra 226.03	Ac 227.03	Th 232.04	Pa 231.04	U 238.03	Np 237.05	Pu 244.06	Am 243.06	Cm 247.07	Bk 247.07	Cf 251.08	Es 252.08	Fm 257.10	Mn 258.10	No 259.10	Lr 262.11	
8	Ce 140.12	Pr 140.91	Nd 144.24	Pm 144.91	Sm 150.36	Eu 151.96	Gd 157.25	Tb 158.93	Dy 162.50	Ho 164.93	Er 167.26	Tm 168.93	Yb 173.05	Lu 174.97				
9	Th 232.04	Pa 231.04	Np 237.05	Pu 244.06	Am 243.06	Cm 247.07	Bk 247.07	Cf 251.08	Es 252.08	Fm 257.10	Mn 258.10	No 259.10	Lr 262.11					

The only metal that is liquid at room temperature.

Period 3
element with
7 valence
electrons.

Semi-conductor
metalloid used
in electronics.

Only one valence electron, but not part of a family.

Extremely reactive non-metal.

Group that will form 2+ cations.

The only non-metal that is liquid at room temperature.

Completely
un-reactive family
with complete
valence shell.

Extremely reactive metal.

Group with 4 valence electrons.

Periodic Trends Review

Dr. Abhishek M



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element chemistry Corrections to Volume I Some comments on notations and terminology List of acronyms and symbols Bibliography

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medicine the environment and biological topics **Relativistic Theory of Atoms and Molecules** Pekka Pyykkö,2012-12-06 **Kaplan SAT Subject Test Chemistry 2015-2016** Kaplan Test Prep,2015-03-03 Essential strategies practice and review to ace the SAT Subject Test Chemistry Getting into a top college has never been more difficult Students need to distinguish themselves from the crowd and scoring well on a SAT Subject Test gives students a competitive edge Kaplan s SAT Subject Test Chemistry is the most up to date guide on the market with complete coverage of both the content review and strategies students need for success on test day Kaplan s SAT Subject Test Chemistry features A full length diagnostic test Full length practice tests Focused chapter summaries highlights and quizzes Detailed answer explanations Proven score raising strategies End of chapter quizzes Kaplan is serious about raising students scores we guarantee students will get a higher score **Merrill Chemistry** Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 **Fundamental World of Quantum Chemistry** Erkki Brändas,Eugene S. Kryachko,2003 Per Olov Löwdin s stature has been a symbol of the world of quantum theory during the past five decades through his basic contributions to the development of the conceptual framework of Quantum Chemistry and introduction of the fundamental concepts through a staggering number of regular summer schools winter institutes innumerable lectures at Uppsala Gainesville and elsewhere and Sanibel Symposia by founding the International Journal of Quantum Chemistry and Advances in Quantum Chemistry and through his vision of the possible and his optimism for the future which has inspired generations of physicists chemists mathematicians and biologists to devote their lives to molecular electronic theory and dynamics solid state and quantum biology Fundamental World of Quantum Chemistry Volumes I II and III form a collection of papers dedicated to the memory of Per Olov Löwdin These volumes are of interest to a broad audience of quantum theoretical physical biological and computational chemists atomic molecular and condensed matter physicists biophysicists mathematicians working in many body theory and historians and philosophers of natural science **The Ultimate Study Guide to Student Success** Dr. Abhishek M,2025-07-30 Success in school and life isn t just about studying harder it s about studying smarter with purpose passion and the right mindset The Ultimate Study Guide for Student Success is more than just a toolkit of strategies it s your personal roadmap to becoming a confident motivated and resilient learner From crafting the perfect study environment to mastering memory techniques managing time like a pro and keeping stress in check this guide equips you with practical tools that are rooted in science and real life application But it doesn t stop there True success also comes from within this book dives deep into the emotional and psychological foundations of learning You ll explore how to overcome procrastination build grit stay focused and cultivate an unstoppable mindset Whether you re a student aiming for top grades or a lifelong learner seeking growth this book will transform not only how you learn but who you become in the process Get ready to unlock your full potential The journey to lasting student success starts here **Chemistry: The Central Science** Theodore L. Brown,H. Eugene LeMay Jr.,Bruce E. Bursten,Catherine Murphy,Patrick Woodward,Steven Langford,Dalius Sagatys,Adrian George,2013-10-04 If you think you

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