

Periodic Trends Questions

5. An element with high ionization energy would most likely be?
- a. A nonmetal with low electronegativity
 - b. A nonmetal with high electronegativity
 - c. A metal with low electronegativity
 - d. A metal with high electronegativity
6. What happens to S when it becomes S^{2-} ?
- a. It loses two electrons and the radius increases
 - b. It loses two electrons and the radius decreases
 - c. It gains two electrons and the radius increases
 - d. It gains two electrons and the radius decreases

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