

100. Ionization energies allow chemists to determine chemical structures. The first two ionization energies of rubidium and strontium are given in the table.

Element	First ionization energy (kJ/mol)	Second ionization energy (kJ/mol)
Rubidium	403	2633
Strontium	550	1064

101. Write an equation to represent the second ionization energy of strontium.

Include state symbols.

_____ [2]

102. Why is the first ionization energy of rubidium larger than the first ionization energy of calcium?

In your answer you should use appropriate chemical terms specified correctly.

_____ [2]

103. Why is the second ionization energy of rubidium larger than the second ionization energy of strontium?

_____ [2]

(Total: 12)

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