

3. For the following, complete with the appropriate information as indicated in parentheses.

Name of compound	Cation	Anion	Chemical formula
tin(II)oxide oxide	Sn^{2+}	O^{2-}	Sn_2O_2
lead(II) bromide	Pb^{2+}	Br^{-}	PbBr_2
tin(IV) sulfide	Sn^{4+}	S^{2-}	Sn_2S_3
nitrogen trifluoride	NF_3	F^{-}	NF_3
manganese(IV) sulfide	Mn^{4+}	S^{2-}	MnS_2

4. Give a correct written explanation as to why they have been named incorrectly and then create the correct name.

Formula	Incorrect name	Explanation	Correct name
SnF_4	stannous fluoride	metal and non-metal $\therefore$ no prefixes	tin(IV) fluoride
NaCl	sodium chloride	sodium is not a multivalent $\therefore$ no prefix	sodium chloride
CuCl_2	copper chloride	copper is multivalent $\therefore$ needs roman numeral	copper (II) chloride
N_2O_5	nitrogen oxide	non metal and non-metal $\therefore$ needs prefixes	dinitrogen pentoxide

5. Classify each of the following as an ionic compound, a molecular compound, or an acid. Write the formula for each one.

	Type	Formula
a) solid sodium sulfate	ionic	Na_2SO_4 (s)
b) aqueous hydrogen nitrate	acid	HNO_3 (aq)
c) gaseous sulfur trioxide	molecular	SO_3 (g)
d) gaseous nitrogen trioxide	molecular	N_2O_3 (g)
e) solid manganese(IV) bromide	ionic	MnBr_2 (s)

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B Lingard



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