

POGIL: Kinetic Molecular Theory

Learning Objectives

- ✓ Identify the basic differences between particle behavior in the solid, liquid, and gaseous phases.
- ✓ Develop an understanding of the postulates of the kinetic molecular theory.
- ✓ Identify how temperature affects molecular motion.
- ✓ Apply the kinetic molecular theory to predict the outcome of everyday situations.

Model 1: Representation of Atoms in Different Phases

The following figures show the representative particles of a given substance in various states of matter.



Source: http://itl.chem.ufl.edu/2045_s00/lectures/lec_f.html

Note: Give complete answers to the key questions, exercises, and problems. Provide explanations and show work, where necessary.

Key Questions

1. What are the key characteristics of atoms and molecules in gases, liquids, and solids? In *Table 1* below, describe the characteristics of particles for each phase of matter based on Model 1. Be specific with regard to spacing, the potential of particles movement, and whether or not the particles will fill the container.

Table 1: Characteristics of the Phases of Matter

	SOLID	LIQUID	GAS
SPACING AND RELATIVE DENSITY			
MOVEMENT & RELATIVE KINETIC ENERGY			
FILLING A CONTAINER (VOLUME)			
"SHAPE" AND MASS AND MISC			

Pogil Kinetic Molecular Activity Answers

Christian Drosten



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Science Citation Index, 1992 Vols for 1964 have guides and journal lists
Proteus Steinmetz, American Electrochemical Society. Meeting, 1908*

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