

Periodic and Groups: Valence Electrons: Mastery

Periodic and Groups

Elements that have the same number of valence electrons have similar properties, so we put them in groups. The the periodic table groups + columns that look like this. They will see family groups 1A-18A. Hydrogen is in the same period as Helium.

Elements that have electrons in the outer shell have similar properties in the Periodic Table. Periods go left to right. Elements in the same vertical columns

Period	Group		Group										18A	
	1A	2A	3A	4A	5A	6A	7A	8A	9A	10A	11A	12A	13A	14A
1	H	He												
2	Li	Be												
3	Na	Mg												
4	K	Ca												

Thinking like the game "Monopoly", answer that an element is in period 3 and group 1.

What element is in Period 2, and Group 16A? (showing 2 means 2, Group 16A is Oxygen 16A. The element is oxygen.)

What element is in Period 5 and Group 1?

What element is in Group 2A, and Period 3?

What group would you find a Halogen in?

What group Magnesium?

Valence Electrons

Valence Electrons are the most important electrons in an atom. Each group contains has the same number of valence electrons. Valence electrons are the electrons that get involved in chemical bonding.

Hydrogen 1



1 valence electron

Helium 2



2 valence electrons

Lithium 3



3 valence electrons

Hydrogen and Helium are both in Group 1A (Hydrogen 1 and Helium 2).

Group 1A, including H and He, has valence electrons. How many valence electrons will they have?

How about Lithium?

Group 18A: - Shows you more noble than have a full shell of electrons. Elements are Helium 2 and Neon 10. Helium is 2 and Neon is 10. Helium and Neon are 18 valence electrons. "He 2 is, Neon 10" Group 18A has a full shell of valence electrons (noble). 18 valence electrons will have, gas, or noble or inert properties.

Periods And Groups Valence Electrons Masses Answers

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Periods And Groups Valence Electrons Masses Answers:

Unveiling the Magic of Words: A Report on "**Periods And Groups Valence Electrons Masses Answers**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Periods And Groups Valence Electrons Masses Answers**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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