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THE PERIODIC TABLE

Chapter 6 Review Package - Part 1

A. Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

	Column A		Column B
37	metal	46.	positively charged ion
2.	halogen	b.	an element whose highest occupied s or ρ sublevels are partially filled
3	inner transition metal	¢.	the energy needed to remove an electron from an atom in the gaseous state
4.	representative element	d.	good conductor of heat and electric current
5	ionization energy	e.	an element whose highest occupied s sublevel and a nearby d sublevel contain electrons
6.	alkaline earth metal	ď.	any nonmetal in Group 7A
7.	cation	g.	the tendency of an atom to attract electrons when the atom is in a compound
8.	electronegativity	h.	an element in which the highest occupied s and ρ sublevels are filled
9.	transition metal	1.0	an element in Group 2A
10.	noble gas	J.	an element whose highest occupied s sublevel and a nearby f sublevel contain electrons

- 11. In the periodic table, there is a periodic pattern in the physical and chemical properties of elements when they are arranged in order of a. increasing atomic mass.
 - b. increasing electronegativity.
 - e. increasing atomic radius.

 - d. increasing atomic number.

Periodic Trends Ch 6 Answers

Heather Hattori, Richard H. Langley

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