

Trends on the Periodic Table

A Periodic Properties Graphing Lab

The periodic table is displayed with colored blocks and group/period labels. The groups are labeled at the top: IA (yellow), IIA (purple), IIIA (purple), IVA (green), VA (green), VIA (green), VIIA (green), VIIIA (green), and VIIIA (green). The periods are labeled on the left: 1 (yellow), 2 (orange), 3 (orange), 4 (orange), 5 (orange), 6 (orange), and 7 (orange). The elements are arranged in rows and columns, with the lanthanide and actinide series shown as separate blocks at the bottom.

Period	Group	Element
1	IA	H
2	IA	Li
2	IIA	Be
3	IA	Na
3	IIA	Mg
4	IA	K
4	IIA	Ca
4	IIIB	Sc
4	IVB	Ti
4	VB	V
4	VIB	Cr
4	VII	Mn
4	VIII	Fe
4	VIII	Co
4	VIII	Ni
4	VIII	Cu
4	IB	Zn
4	IIA	Ga
4	IIIA	Ge
4	IVA	As
4	VA	Se
4	VIA	Br
4	VIIA	Kr
5	IA	Rb
5	IIA	Sr
5	IIIB	Y
5	IVB	Zr
5	VB	Nb
5	VIB	Mo
5	VII	Tc
5	VIII	Ru
5	VIII	Rh
5	VIII	Pd
5	VIII	Ag
5	IB	Cd
5	IIA	In
5	IIIA	Sn
5	IVA	Sb
5	VA	Te
5	VIA	I
5	VIIA	Xe
6	IA	Cs
6	IIA	Ba
6	IIIB	La
6	IVB	Hf
6	VB	Ta
6	VIB	W
6	VII	Re
6	VIII	Os
6	VIII	Ir
6	VIII	Pt
6	VIII	Au
6	IB	Hg
6	IIA	Tl
6	IIIA	Pb
6	IVA	Bi
6	VA	Po
6	VIA	At
6	VIIA	Rn
7	IA	Fr
7	IIA	Ra
7	IIIB	Ac
7	IVB	Db
7	VB	Rf
7	VIB	Bh
7	VII	Hs
7	VIII	Mt
7	VIII	Ds
7	VIII	Uus
7	IB	Uub
7	IIA	Uuh
7	IIIA	Uug
7	IVA	Uub
7	VIA	Uuo

The lanthanide series (f-block) is shown below the main table, containing elements Ce, Pr, Nd, Pm, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, and Lu. The actinide series (f-block) is shown below the lanthanide series, containing elements Th, Pa, U, Np, Pu, Am, Cm, Bk, Cf, Es, Fm, Md, and No.



Physical Science and Chemistry, Grades 9 - 12

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