

Oxidation and Reduction

When copper(II) nitrate is transformed in a chemical reaction?

Q1a)?

Below reactions show the process in copper(II) nitrate transformation in the air. Copper gets oxidized in chemical gases present in the air. Metals react or combine in the presence of air and water. Nitrogenic gases responsible for the air decomposition leads the air of the nitric oxide present gas. Nitric acid, ammonia, H_2O and other substances are examples of oxidation and reduction, nitrogenic bacteria in various reactions. In this activity you will explore when is happening in the nitric acid in various reactions.

Model 1 – Redox Reactions

Redox Reactions:



Redox Reactions:



1. What are typical reactions in Model 1?

Redox reactions and reduction reactions.

2. Explain the redox and reduction reactions in Model 1. What does the transformation in the redox reactions that would allow you to identify them as redox reactions? If you can specify, describe how Model 1 is supported your answer.

There is no reaction showing the transformation of the redox reactions.

Pogil Lab Oxidation Answer Key

Kathleen Armour



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