

Oxidation and Reduction

When happens when elements are transformed in a chemical reaction?

What?

Below reactions show the process by which iron is either transformed in the air. Copper gets oxidized in dissolved gases present in water. Metals react or corrode in the presence of dissolved water. Sulfuric acid is responsible for corrosion in structures for transportation and also in other structures as pipelines from remote oil refineries. All of these reactions are examples of oxidation and reduction, otherwise known as redox reactions. In this section you will explore when is happening in the above listed in redox reactions.

Model 1 – Redox Reactions

Redox Reactions:



Redox Reactions:



1. What are typical reactions you observe in Model 1?

Redox reactions and redox reactions.

2. Explain the redox and reduction reactions in Model 1. What does each reaction tell you about the reaction? How does Model 1 use copper/copper ions?

There is no reaction showing the reaction between the two redox reactions.

Pogil Chemistry Answer Key Oxidation And Activities

Clifford Lane



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Oxidation and Reduction in Inorganic and Analytical Chemistry Alan Vincent, 1985-11-22 Beginning with basic principles this self instructional text leads students to an advanced understanding of oxidation and reduction Covers use of the mole concept and equation balancing and elementary thermodynamics concepts in a clear accessible manner Self tests and revision notes follow each chapter **Oxidation and Reduction** ,1992 *Oxidation-reduction Reactions* Irish Science Teachers' Association. Chemistry Syllabus Committee, 1988* Keys to Oxidation-reduction ,1974

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