Period Date Tuesday, November 1 2016 Name

# Periodic Trends

#### ATOMIC RADIUS

What trend in atomic radius do you see as you go down a group/family on the periodic table?

2. What causes this trend?

3. What trend in atomic radius do you see as you go across a period/row on the periodic table?

4. What causes this trend?

Circle the atom in each pair that has the largest atomic radius.

a) Al B

b) S O c) Br Cl

d) Na Al e) O F f) Mg Ca

Put the following elements in order from smallest to largest atomic radius and explain why: C. O. Sn. Sr.

#### ELECTRONEGATIVITY

Define electronegativity

8. How does the ionic radius of a nonmetal compare with its atomic radius?

9. What trend in electronegativity do you see as you go down a group/family on the periodic table?

10. What causes this trend?

11. What trend in electronegativity do you see as you go across a period/row on the periodic table?

12. What causes this trend?

Circle the atom in each pair that has the greater electronegativity.

a) Ca Ga b) Li O c) Cl S d) Br As

e) Ba Sr

f) O S

**GENERAL QUESTIONS** 

14. Which group tends to form +1 ions?

15. Which group tends to form +2 ions?

16. Which group tends to form -1 ions?

17. Which group tends not to form ions or react? \_\_\_\_\_

# **Periodic Trends Quiz Answer Key**

R Bogdan

**Periodic Trends Quiz Answer Key:** 

Unveiling the Magic of Words: A Review of "Periodic Trends Quiz Answer Key"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Periodic Trends Quiz Answer Key**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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