

Name: \_\_\_\_\_

Period: \_\_\_\_\_

## Periodic Table WebQuest

**Instructions:** In each of the websites listed, find the information asked for.



### Part I: "Get Organized Periodically"

<http://www.chemicalsciences.com/lesson/periodic.html>

1. Why are the elements placed in specific places on the Periodic Table?
2. Periods are \_\_\_\_\_ that run from left to right.
3. Elements in the same period have the same \_\_\_\_\_.
4. Every element in the first period has \_\_\_\_\_ shell for its \_\_\_\_\_. Every element in the second period has \_\_\_\_\_ for its \_\_\_\_\_. See the pattern?
5. Groups are \_\_\_\_\_ that run from top to bottom.
6. The elements of a group have the same number of \_\_\_\_\_ in their \_\_\_\_\_ shell.
7. Every element in group one has \_\_\_\_\_ electron in its outer shell. Every element in group two has \_\_\_\_\_ electrons in its outer shell.
8. Hydrogen is special because it can act like two groups, \_\_\_\_\_ and \_\_\_\_\_.
9. Hydrogen sometimes is \_\_\_\_\_ an electron and sometimes it has an \_\_\_\_\_ electron.
10. Although helium has only \_\_\_\_\_ electrons in its outer shell, it is grouped with elements that have \_\_\_\_\_.
11. The green elements on this table are called \_\_\_\_\_ elements. They each have two electrons in their outer shell.

### Part II: "Family Fun" Go to <http://chemicalsciences.com/>

12. Click on Alkali Metals (left bar) and answer the following questions.
  - a. What is the group number? \_\_\_\_\_
  - b. Are these metals reactive? \_\_\_\_\_
  - c. Do these metals occur freely in nature? \_\_\_\_\_
  - d. How many electrons are in their outer shell? \_\_\_\_\_

# Periodic Table Webquest Answer Key

**A Gutmann**



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