

s- block		2.7 Table : Modern Periodic Table																p- block	
		Atomic No. Symbol Name Atomic mass																	
1 H Hydrogen 1.008	2 He Helium 4.003																	10 Ne Neon 20.18	18 Ar Argon 39.95
3 Li Lithium 6.941	4 Be Beryllium 9.012																	19 K Potassium 39.10	36 Kr Krypton 83.80
11 Na Sodium 22.99	12 Mg Magnesium 24.31																	35 Br Bromine 79.90	54 Xe Xenon 131.3
19 K Potassium 39.10	20 Ca Calcium 40.08	21 Sc Scandium 44.96	22 Ti Titanium 47.88	23 V Vanadium 50.94	24 Cr Chromium 52.00	25 Mn Manganese 54.94	26 Fe Iron 55.85	27 Co Cobalt 58.93	28 Ni Nickel 58.69	29 Cu Copper 63.55	30 Zn Zinc 65.38	31 Ga Gallium 69.72	32 Ge Germanium 72.64	33 As Arsenic 74.92	34 Se Selenium 78.96	35 Br Bromine 79.90	36 Kr Krypton 83.80	53 I Iodine 126.9	86 Rn Radon 222
37 Rb Rubidium 85.47	38 Sr Strontium 87.62	39 Y Yttrium 88.91	40 Zr Zirconium 91.22	41 Nb Niobium 92.91	42 Mo Molybdenum 95.94	43 Tc Technetium 98.91	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.36	47 Ag Silver 107.87	48 Cd Cadmium 112.41	49 In Indium 114.82	50 Sn Tin 118.71	51 Sb Antimony 121.76	52 Te Tellurium 127.6	53 I Iodine 126.9	54 Xe Xenon 131.3	85 At Astatine 210	118 Og Oganesson 284
55 Cs Cesium 132.91	56 Ba Barium 137.33	57-71 Lanthanides and Actinides	72 Hf Hafnium 178.49	73 Ta Tantalum 180.95	74 W Tungsten 183.84	75 Re Rhenium 186.21	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.08	79 Au Gold 196.97	80 Hg Mercury 200.59	81 Tl Thallium 204.38	82 Pb Lead 207.2	83 Bi Bismuth 208.98	84 Po Polonium 209	85 At Astatine 210	86 Rn Radon 222	117 Ts Tennessine 289	119 Nh Nihonium 284
87 Fr Francium 223	88 Ra Radium 226	89-103 Lanthanides and Actinides	104 Rf Rutherfordium 261	105 Db Dubnium 262	106 Sg Seaborgium 266	107 Bh Bohrium 264	108 Hs Hassium 277	109 Mt Meitnerium 268	110 Ds Darmstadtium 271	111 Rg Roentgenium 272	112 Cn Copernicium 285	113 Nh Nihonium 284	114 Fl Flerovium 289	115 Mc Moscovium 288	116 Lv Livermorium 293	117 Ts Tennessine 289	118 Og Oganesson 284	119 Nh Nihonium 284	120 Ds Darmstadtium 271
f- block																			
☆		57 La Lanthanum 138.91	58 Ce Cerium 140.12	59 Pr Praseodymium 140.91	60 Nd Neodymium 144.24	61 Pm Promethium 145	62 Sm Samarium 150.36	63 Eu Europium 151.96	64 Gd Gadolinium 157.25	65 Tb Terbium 158.93	66 Dy Dysprosium 162.50	67 Ho Holmium 164.93	68 Er Erbium 167.26	69 Tm Thulium 168.93	70 Yb Ytterbium 173.05	71 Lu Lutetium 174.97			
#		89 Ac Actinium 227	90 Th Thorium 232.04	91 Pa Protactinium 231.04	92 U Uranium 238.03	93 Np Neptunium 237.05	94 Pu Plutonium 244	95 Am Americium 243	96 Cm Curium 247	97 Bk Berkelium 247	98 Cf Californium 251	99 Es Einsteinium 252	100 Fm Fermium 257	101 Md Mendelevium 258	102 No Nobelium 259	103 Lr Lawrencium 262			

## Periods and electronic configuration



Can you tell?

- On going through the modern periodic table it is seen that the elements Li, Be, B, C, N, O, F and Ne belong to the period-2. Write down electronic configuration of all of them.
- Is the number of valence electrons same for all these elements?
- Is the number of shells the same in these ?

You will find that the number of valence electrons is different in these elements. However, the number of shells is the same. You will also find that, while going from left to right, within the period, the atomic number increases by one at a time and the number of valence electrons also increases by one at a time.

	1	2	13	14	15	16	17	18
1	H 1							He 2
2	Li 2,1	Be 2,2	B 2,3	C 2,4	N 2,5	O 2,6	F 2,7	Ne 2,8
3	Na 2,8,1	Mg 2,8,2	Al 2,8,3	Si 2,8,4	P 2,8,5	S 2,8,6	Cl 2,8,7	Ar 2,8,8
4	K 2,8,8,1	Ca 2,8,8,2						
5		Sr						
6		Ba						
7		Ra						

Potassium atom

Argon atom

## 2.8 New period new shell

# Periodic Table Physical Science If8767 Answers

**D. Michael P. Mingos**



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## **Table of Contents Periodic Table Physical Science If8767 Answers**

1. Understanding the eBook Periodic Table Physical Science If8767 Answers
  - The Rise of Digital Reading Periodic Table Physical Science If8767 Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Physical Science If8767 Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Physical Science If8767 Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Physical Science If8767 Answers
  - Personalized Recommendations
  - Periodic Table Physical Science If8767 Answers User Reviews and Ratings
  - Periodic Table Physical Science If8767 Answers and Bestseller Lists
5. Accessing Periodic Table Physical Science If8767 Answers Free and Paid eBooks
  - Periodic Table Physical Science If8767 Answers Public Domain eBooks
  - Periodic Table Physical Science If8767 Answers eBook Subscription Services
  - Periodic Table Physical Science If8767 Answers Budget-Friendly Options
6. Navigating Periodic Table Physical Science If8767 Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Periodic Table Physical Science If8767 Answers Compatibility with Devices
  - Periodic Table Physical Science If8767 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Table Physical Science If8767 Answers
  - Highlighting and Note-Taking Periodic Table Physical Science If8767 Answers
  - Interactive Elements Periodic Table Physical Science If8767 Answers

8. Staying Engaged with Periodic Table Physical Science If8767 Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Table Physical Science If8767 Answers
9. Balancing eBooks and Physical Books Periodic Table Physical Science If8767 Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table Physical Science If8767 Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Physical Science If8767 Answers
  - Setting Reading Goals Periodic Table Physical Science If8767 Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Physical Science If8767 Answers
  - Fact-Checking eBook Content of Periodic Table Physical Science If8767 Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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web restful web services rest stands for representational state transfer rest is an architectural style not a protocol

advantages of restful web services fast restful web services are fast because there is no strict specification like soap it consumes less bandwidth and resource

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