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Exceed to Succeed

Worksheet: Periodic Trends

1. ATOMIC RADIUS

For each of the following sets of atoms, rank the atoms from smallest to largest atomic radius.

- a. Li, C, F
- b. Li, Na, K
- c. Ge, P, O
- d. C, N, Al
- e. Al, Cl, Ga

2. IONIC RADIUS

For each of the following sets of ions, rank them from smallest to largest ionic radius.

- a. Mg^{2+} , Si^{4+} , S^{2-}
- b. Mg^{2+} , Ca^{2+} , Ba^{2+}
- c. F^- , Cl^- , Br^-
- d. Ba^{2+} , Cu^{2+} , Zn^{2+}
- e. Si^{4+} , P^{3-} , O^{2-}

3. IONIZATION ENERGY

For each of the following sets of atoms, rank them from lowest to highest ionization energy.

- a. Mg, Si, S
- b. Mg, Ca, Ba
- c. F, Cl, Br
- d. Ba, Cu, Ne
- e. Si, P, He

4. ELECTRONEGATIVITY

For each of the following sets of atoms, rank them from lowest to highest electronegativity.

- a. Li, C, N
- b. C, O, Ne
- c. Si, P, O
- d. K, Mg, P
- e. S, F, He

Periodic Trends Chapter 5 Answer Key

S Marginson



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application process Understanding the Exam Pattern Sections in AP ECET Marking scheme Time management tips Detailed Syllabus Breakdown Mathematics Physics Chemistry Engineering Branch specific topics Civil Mechanical Electrical Electronics Computer Science Study Strategies and Planning How to create a study timetable Daily and weekly goals Time management techniques Dealing with exam stress Topic wise Study Material Mathematics Key concepts and formulas Physics Important principles and applications Chemistry Essential topics and reaction mechanisms Branch specific topics Important points and problem solving techniques Practice Questions and Model Papers Previous year questions Mock tests Solved examples Tips and Tricks for Exam Day How to attempt questions Guessing strategies Time allocation per section Post Exam Guidance Result analysis Counseling process Admission procedures **Chemistry** Steven S. Zumdahl,1986

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