

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular C_2H_6O (molecules)

Formula represents the number of _____ of each
in a single molecule.

Empirical C_2H_6O (molecules)

Formula represents the simplest ratio of
positive and negative ions in a formula
unit.

Atomic Mass

The average mass of an atom, weighted
average.

Isotopes

^{12}C and ^{13}C isotopes

Same protons but different
neutrons.

Abundance

^{12}C is abundant.

Same protons but different

^{13}C is less

Note Taking Guide Episode 6chemical Formula

A Gutmann



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