

Extra Testing Section - (Experiments 1,2,3) - Part II

Evidence for wave nature of light:

1. Interference experiments have revealed that light exhibits wave-like behavior in situations:
 - a. double-slit interference
 - b. thin film interference
 - c. diffraction
2. Huygens' principle can explain interference. Huygens is formulated as: a point on a wave behaves as if it were a source of spherical waves.
3. Light exhibits particle-like behavior in how it interacts with matter. For example, how it is absorbed and re-emitted by atoms and molecules and how it interacts with matter in a vacuum.

Evidence for particle nature of light:

1. Photoelectric effect: light propagates as waves
 - a. Without sufficient frequencies of light, electrons are not ejected from a metal surface.
 - b. Energy transfers as particles called photons.
 - c. Electrons transfer the energy of photons to eject electrons from a metal surface.
 - d. Energy transfers as particles called photons.
2. Compton Scattering results in photon.

Student should fill light <u>propagates</u> <u>as waves</u> and <u>interacts</u> <u>as particles</u> and <u>energy</u> <u>transfers</u> <u>as particles</u> called <u>photons</u>.
Without this photoelectric effect, electrons in atoms are <u>not</u> <u>ejected</u>. There is transfer of <u>energy</u> <u>from</u> <u>light</u> <u>to</u> <u>electrons</u> <u>in</u> <u>atoms</u>.
Light light <u>propagates</u> <u>as</u> <u>waves</u> <u>and</u> <u>interacts</u> <u>as</u> <u>particles</u> <u>called</u> <u>photons</u> <u>and</u> <u>energy</u> <u>transfers</u> <u>as</u> <u>particles</u> <u>called</u> <u>photons</u> <u>and</u> <u>energy</u> <u>transfers</u> <u>as</u> <u>particles</u> <u>called</u> <u>photons</u>.

3. The wave theory cannot explain the photoelectric effect.

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