

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ behavioral _____ approaches _____

2. $40g = 40r + 4 \cdot 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide.

The _____ is a _____ reaction _____ give the _____ _____ _____ from the _____ reaction in the _____

3. Problem

Given a chemical substance reacting with a chemical system, determine whether it produces

$$2 \cdot A_2 + 3g \cdot B \rightarrow 2AB_2$$

make reaction

$$\begin{array}{cccc} 2A & A_2 & B & B_2 \\ 2A & A_2 & B & AB_2 \\ 2A & B_2 & B & AB_2 \end{array}$$

2. If you start with 4 moles of A_2 , how many moles of AB_2 will be produced?

$$2 \cdot 4A_2 \rightarrow 2AB_2 \cdot 1 \text{ mol } B_2 \rightarrow 2 \text{ mol } AB_2$$

Note Taking Guide Episode 80stoichiometry Answers

JG Myers



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