Quantum Numbers

- Used to an in an
 - 1
 - Represents ______ energy level of _______

_____ ar of _____ in an

Example: What is the maximum number of electrons that can be in the _____ main energy level?

£

- The ______
- Describes the ______ within an ______
- of arbital _____possible is ______

Orbital Shopes

designated_____

- level 21
- level 3
- level 4:

How many electrons can each sublevel hold?

- s = 1 orbital x 2 e /orbital = ____e
- p = 3 orbitals x 2 e /orbital = ____e
- d = 5 orbitals x 2 e /orbital = ____e
 - f = 7 orbitals x 2 e /orbital = ____e

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A Loxley

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