

Note Taking Guide: Episode 301 Name_____

Model: A familiar idea used to explain unfamiliar facts observed in nature.

Theory: An explanation of observable facts and phenomena. To

remain valid, models and theories must:

- explain all known facts
- enable scientists to make correct predictions.

Democritus:

- proposed the existence of an atom
- word comes from the greek word atomis which means not to cut or "indivisible"

Aristotle

- rejected the idea of the atom
- said matter could be cut continually

Dalton's theory proposed that atoms:

- are the building blocks of matter
- are indivisible
- of the same elements are identical
- of different elements are different
- unite in small, whole number ratios to form compounds.

J.J. Thomson

- credited with discovery of electron: a blow to Dalton's indivisible atom.
- proposed the plum pudding model of the atom:
Negatively charged electrons embedded in a ball of positive charge

Note Taking Guide Episode 301 Chemistry

William Swan Sonnenschein



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Industrial & Engineering Chemistry ,1923 The Journal of Industrial and Engineering Chemistry ,1923 **The Chemical Bulletin** ,1916 **The Chicago Chemical Bulletin** ,1914 *Books of 1912-* ,1922 **A Reader's Guide to Contemporary Literature** William Swan Sonnenschein,1896 *Books of 1921-1925* Chicago Public Library,1927
Environmental Health Perspectives ,1993 **Scientific American** ,1900 Monthly magazine devoted to topics of general scientific interest **Readers' Guide to Periodical Literature** Anna Lorraine Guthrie,Bertha Tannehill,Neltje Marie Tannehill Shimer,1919 An author subject index to selected general interest periodicals of reference value in libraries

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