# Polyatomic Ions Chart

Formula	Name	Formula	Name
NH4°	Ammonium	CrO <sub>4</sub> -2	Chromate
NH <sub>3</sub>	Ammonia	Cr2O7-2	Dichromate
C2H3O2	Acetate	MnO <sub>4</sub>	Permanganate
CH₃COO-	Acetate	MnO <sub>4</sub> -2	Manganate
CN-	Cyanide	NO <sub>2</sub>	Nitrite
CO3-2	Carbonate	NO₃ <sup>-</sup>	Nitrate
HCO3	Bicarbonate	OH-	Hydroxide
C2O4-2	Oxalate	PO <sub>4</sub> -3	Phosphate
CIO	Hypochlorite	SCN-	Thiocyanate
CIO2-	Chlorite	Fe(CN) <sub>6</sub> -3	Ferricyanide
CIO3	Chlorate	5O₃ <sup>-2</sup>	Sulfite
CIO4	Perchlorate	5O <sub>4</sub> -2	Sulfate
5 <sub>2</sub> O <sub>3</sub> -2	Thiosulfate	H5O <sub>4</sub> -	Hydrogen sulfate
BrO-	Hypobromite	IO3-	Iodate
AsO2-3	Arsenite	SeO <sub>4</sub> -2	Selenate
BrO <sub>3</sub>	Bromate	HSO₃⁻	Hydrogen sulfite

# **Polatomic Compounds Study Guide Answers**

**M Tight** 

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