Unit 1: Matter and Energy

Name: KEY
Date: Class:

<u>Directions</u>: Complete the following review questions. Review your Cornell notes, learning targets, worksheets, labs, and other assignments to adequately prepare for the unit test.

Learning Target(s): 1, 2, 3, 4, 10

- In the Melting of a Solid lab, you heated (or cooled) an unknown solid and recorded its temperature at 30-second intervals. From this data you created a heating (or cooling) curve to identify the substance.
 - a. Identify the independent variable of this experiment: the time you heat the solid for
 - b. Identify the dependent variable of this experiment: the temperature of the solid
 - c. Identify the control for this experiment; substance used; method of heating
- 2. Decide whether the following observations are qualitative or quantitative:
 - The solution began to get cloudy once the second reagent was added. Qualitative
 - The piece of magnesium was 1.25 centimeters in length. Quantitative
 - c. The mass of the block decreased by fifteen grams. Quantitative
 - d. The test tube appeared to have a dark orange solid suspended in the solution. Qualitative
 - e. The residue left in test tube 5 was slightly larger than in test tube 4. Qualitative

Learning Torget(s): 15

- 3. Suggest a process to use to separate each of the following mixtures:
 - a. A mix of salt and water:
 - b. A smoothie drink with veggies and fruits of different densities:
- 4. Which physical property does filtration use to separate a mixture?

Learning Target(s): 5, 6

- 5. Describe at least two safety precautions to use when using a Bunsen Burner in lab:
 - a. Tie back hair and loose clothing
 - b. Light match first then turn on gas valve
- Which type of lab equipment would be most appropriate to measure 4 mL of water in? Choose one and explain why.
 - n. A 250 mL beaker

d. A 50 mL graduated cylinder

b. A 10 mL graduated cylinder

e. A large test tube

c. An electronic balance

f. A thermometer

You would get the most accurate measurement with this size. This piece of equipment has graduations to measure volume.

Physical Science Chemistry Unit Review Guide Answers

Lauren Gardner

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Chemical News and Journal of Physical Science William Crookes, James H. Gardiner, Gerald Druce, H. W.

Blood-Ryan, 1865 The Chemical News and Journal of Physical Science, 1865 The Literary Guide and Rationalist

Review, 1925 Basic Concepts of Chemistry, Study Guide Leo J. Malone, 2000 The text's three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the

math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems El-Hi Textbooks in Print ,1982 El-Hi Textbooks & Serials in Print , 2003 ,2003

Basic Physical Science Helen Dolman MacCracken,1964 ERIC Educational Documents Index, 1966-69: Major descriptors ,1970 ERIC Educational Documents Index, 1966-1969: Minor descriptors. Author index CCM Information Corporation,1970 Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff,2003-03

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