

Periodic Trends Worksheet

Using a few online sources (including textbooks), answer the following questions. Please provide a brief explanation of each answer.

Additional help: use the periodic trends/properties as clues you've learned in this video: <https://www.youtube.com/watch?v=0mD8b6-2G0g>

1. What elements would be most likely to be trading electrons (oxidation is largest) with lithium, aluminum, and sodium?
2. Magnesium and calcium have a larger atomic radius than aluminum.
3. Inductively electronegative will be the largest atomic radii in each pair:
 - a. Na or Li
 - b. Al or Mg
 - c. Cl or C
4. Inductively electronegative will be the largest atomic radii in each pair:
 - a. R or Ca
 - b. Carbon or
 - c. Li or Be
5. Inductively electronegative will be the most electronegative element in each pair:
 - a. Cl or C
 - b. Na or Li
 - c. R or Be
6. Inductively electronegative will be the greatest electronegativity in each pair:
 - a. Al or C
 - b. Al or Si
 - c. R or Be
7. Inductively electronegative will be the greatest electronegativity in each pair:
 - a. R or Be
 - b. C or Ca
 - c. Al or Mg
8. Chlorine is the most electronegative of each of these 4 periodic trends. The most electronegative:
 - a.
 - b.

Periodic Trends Questions And Answers

The Princeton Review



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