

Test Periodic Table and Trends SCH3U

Name: _____

Multiple Choice: Select the most correct answer (20K)

1. The mass number of an atom is determined by:

- a. the number of protons only
- b. adding the neutrons and protons
- c. adding the neutrons and electrons
- d. adding the protons and electrons



2. The letter "X" in the diagram above marks the:

- a. Region of greatest density
- b. Location of the protons
- c. Location of the neutrons
- d. Location of the electrons
- e. Nucleus

3. Most of the mass of the atom can be found in the:

- a. charges b. electron cloud c. electrons d. nucleus

4. Of the basic atomic particles, the one that would be attracted to a negatively charged metallic plate is the:

- a. proton b. neutron c. electron d. none of the above

5. Isotopes (such as hydrogen-1, hydrogen-2, and hydrogen-3) are atoms of the same element that differ in:

- a. the number of protons in the nucleus
- b. the number of electrons
- c. the atomic number
- d. the number of neutrons in the nucleus
- e. the mass number

6. Which of the following accurately summarizes the isotope Argon-40?

- a. 40 protons, 40 electrons, 18 neutrons
- b. 18 protons, 18 electrons, 22 neutrons
- c. 39.95 protons, 39.95 electrons, 21.05 neutrons
- d. 22 protons, 22 electrons, 18 neutrons

Periodic Table Test Answers

M Carnoy



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In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Periodic Table Test Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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