

# Periodic Table Basics

**Step 1: Complete the squares for each element by adding the atomic number, name, and atomic mass.**

Write the atomic number at the top of the square.

Write the element's name under the symbol.

Write the atomic mass at the bottom of the square.

**Step 2: Determine the number of protons, neutrons, and electrons in each element.**


**Step 3: Create a Bohr diagram for each element.**

**Step 4: Draw the Lewis Structure for each element.**


5  
**B**  
 Boron  
 10.81

P =   5    
 N =   6    
 E =   5

Bohr Diagram:



Lewis structure:



**Step 5: Use the following colors to shade in the square for each element. You should ONLY color in the small square in the upper left-hand corner and not the entire card.**

Green = Li & Na

Pink = O & S

Blue = Be & Mg

Purple = F & Cl

Orange = B & Al

Red = C & Si

Tan = N & P

Yellow = He, Ne, & Ar

**Step 6: Cut the cards apart and arrange according to atomic number in the pattern shown below. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.**



**Step 7: Answer the questions on the back of this worksheet using the information on your Periodic Table.**

# Periodic Table Basics Answers

**Heather Hattori, Richard H. Langley**



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## **Table of Contents Periodic Table Basics Answers**

1. Understanding the eBook Periodic Table Basics Answers
  - The Rise of Digital Reading Periodic Table Basics Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Basics Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Basics Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Basics Answers
  - Personalized Recommendations
  - Periodic Table Basics Answers User Reviews and Ratings
  - Periodic Table Basics Answers and Bestseller Lists
5. Accessing Periodic Table Basics Answers Free and Paid eBooks
  - Periodic Table Basics Answers Public Domain eBooks
  - Periodic Table Basics Answers eBook Subscription Services
  - Periodic Table Basics Answers Budget-Friendly Options
6. Navigating Periodic Table Basics Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Periodic Table Basics Answers Compatibility with Devices
  - Periodic Table Basics Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Table Basics Answers
  - Highlighting and Note-Taking Periodic Table Basics Answers
  - Interactive Elements Periodic Table Basics Answers
8. Staying Engaged with Periodic Table Basics Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Periodic Table Basics Answers
- 9. Balancing eBooks and Physical Books Periodic Table Basics Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table Basics Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Basics Answers
  - Setting Reading Goals Periodic Table Basics Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Basics Answers
  - Fact-Checking eBook Content of Periodic Table Basics Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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