

Name _____ Date _____

PHYSICAL & CHEMICAL CHANGE

In a physical change, the original substance still exists; it has only changed in form (i.e. size, shape, state of matter, etc). In addition, a physical change is always reversible - you can get back what you started with! In a chemical change, however, a completely new substance is produced - you can never get back what you started with! Energy (heat) changes typically accompany chemical changes, but can sometimes occur in physical changes as well. Do not assume a chemical change is taking place just because energy is involved! Think: Do I have a totally different substance than I started with? If so, this is always a chemical change.

Read the following statements. Classify the following as being a physical or chemical change.

PHYSICAL CHANGE	CHEMICAL CHANGE	STATEMENTS
		An ice cube melts on a hot summer day.
		Green grass making its own food.
		Hydrochloric acid reacts with potassium hydroxide to produce salt, water, and heat.
		Pancakes cook on a griddle.
		non-metals do not conduct heat or electricity well
		On a cold fall morning, water vapor in the atmosphere condenses into dew on the grass.
		A tire is deflated (the air is let out).
		Potassium chlorate decomposes to potassium chloride and oxygen gas.
		When placed in water, a sodium pellet catches on fire as hydrogen gas is given off and sodium hydroxide forms.
		Wood rots out in the rain.
		An iron nail rusts when left out in the rain.
		A pellet of sodium is sliced in two.
		Food is digested in the stomach
		Milk sours after being kept past the expiration date.
		A pellet of sodium is sliced in two.
		Sugar dissolves in water.
		Sodium hydroxide dissolves in water.

Physical Change Vs Chemical Change Worksheet

Chemistry If8766

Richard G. Cote, Darcy O. Blauvelt



Physical Change Vs Chemical Change Worksheet Chemistry If8766:

Physical and Chemical Changes (eBook) Edward P. Ortleb, Richard Cadice, 1993-09-01 This book presents a program of basic studies in physical and chemical changes of matter. The definition of matter is presented along with explanations of states and properties of matter. Topics include atoms, molecules, elements, compounds, mixtures, solutions, symbols, and formulas. Each of the twelve teaching units in this book is introduced by a color transparency, print books, or PowerPoint slide eBooks that emphasizes the basic concept of the unit and presents questions for discussion. Reproducible student pages provide reinforcement and follow-up activities. The teaching guide offers descriptions of the basic concepts to be presented, background information, suggestions for enrichment activities, and a complete answer key. *Physical vs. Chemical Changes in Matter | Understanding Chemical Properties of Matter | Grade 6-8 Physical Science* Baby Professor, 2024-04-15 Dive into the transformative world of matter with *Physical vs Chemical Changes in Matter*. This essential guide for grades 6-8 demystifies the complexities behind physical and chemical changes. It lays a solid foundation for understanding chemical properties, from melting ice to rusting metal. A must-have for educators, homeschooling parents, and school libraries, it's aligned with the US STEM curriculum, ensuring young scientists grasp these fundamental concepts. Ignite curiosity about the world's material makeup today. [Properties of Matter: Physical Changes vs. Chemical Changes Gr. 5-8](#) George Graybill, 2015-09-01 This is the chapter slice *Physical Changes vs Chemical Changes* from the full lesson plan *Properties of Matter*. Discover what matter is and is not. Learn about and the difference between a mixture and a solution. Chocked full with hands-on activities to understand the various physical and chemical changes to matter. Our resource provides ready-to-use information and activities for remedial students using simplified language and vocabulary. Written to grade, these science concepts are presented in a way that makes them more accessible to students and easier to understand. Our resource is jam-packed with experiments, reading passages, and activities all for students in grades 5 to 8. Color mini posters and answer key included and can be used effectively for test prep and your whole class. All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives. [Chemical Changes](#) Rebecca L. Johnson, 2008-06-01 Everything around us is matter. Physical properties describe what matter is like, and chemical properties describe what matter can do. In a physical change, the physical properties of matter change from one state to another. During a chemical change, the chemical properties change and produce a new substance. Signs of these changes include sound, light, color, smell, or bubbles. [Physical and Chemical Changes in Matter : Chemistry Science Book for Grade 2 | Children's Chemistry Books](#) Baby Professor, 2022-12-01 You are now ready to learn about the physical and chemical changes in matter, which is an introduction to chemistry. The first chapter teaches you about elements or the basic blocks of matter. Next, you will read about what is physical change and its types. The third chapter discusses what is chemical change and its examples. Start reading today.

Chemical Changes Rebecca L. Johnson, 2008-06-01 Matter has both physical and chemical properties. During a chemical

change the chemical properties of a substance change and one or more new substances form Every chemical reaction involves a transfer of energy During a chemical reaction atoms in matter are rearranged but no atoms are gained or lost

Chemical Changes Lynnette Brent, 2008-08-15 An introduction to how chemicals react and change Physical Vs. Chemical Changes in Matter Understanding Chemical Properties of Matter Grade 6-8 Physical Science Baby Professor, 2024-01-04 Dive into the transformative world of matter with *Physical vs Chemical Changes in Matter* This essential guide for grades 6-8 demystifies the complexities behind physical and chemical changes It lays a solid foundation for understanding chemical properties from melting ice to rusting metal A must have for educators homeschooling parents and school libraries it is aligned with the US STEM curriculum ensuring young scientists grasp these fundamental concepts Ignite curiosity about the world's material makeup today Chemical Change Darlene R. Stille, 2005-09 Learn about chemical changes and what causes them **Physical and Chemical Change** Alan V. Jones, Nigel Heslop, Mike Clemmet, 1995 Bad Reaction! Evidence of a Chemical Reaction | Endothermic vs Exothermic Reactions | Grade 6-8 Physical Science Baby Professor, 2024-04-15 Unlock the mysteries of chemical reactions with this engaging book tailored for middle schoolers Perfect for educators homeschooling parents and librarians it delves into the transformative world of endothermic and exothermic reactions showcasing the dynamic nature of chemistry Discover the signs and wonders of chemical changes from fireworks to baking pies emphasizing these reactions critical role in the US STEM curriculum Equip your students with the knowledge to explore the fascinating changes around us **Chemical Changes** Rebecca Kraft Rector, 2019-07-15 A burning log a rotting banana and a rusting bicycle These are examples of chemical changes With this accessible narrative readers are introduced to the basics of chemical changes Easy to follow explanations cover matter forms properties and changes Further detail introduces how to make chemical changes and how to recognize them Color photographs fast facts and an engaging activity expand and reinforce the newly learned ideas This book supports Next Generation Science Standards correlating directly with the NGSS Grade 2 standard to classify different kinds of materials by their observable properties *Understanding Physical and Chemical Changes* Richard G. Cote, Darcy O. Blauvelt, 2012-06 In *Understanding Physical and Chemical Changes* a science unit students work to understand the concepts and definitions of physical and chemical change Grades 6-8 **Chemical Changes**, 2003-01-14 Any change that results in the formation of a new kind of matter is a chemical change Energy is always involved in a chemical change sometimes with spectacular results *Chemical Reactions 6-Pack* Jenna Winterberg, 2015-09-20 Learn about six types of chemical reactions activation energy and hopping electrons reactivity catalysts and inhibitors physical changes of mixtures and more with this high interest nonfiction title This 6 Pack provides five days of standards based activities that will engage fifth grade students support STEM education and build content area literacy in life science It includes vibrant images fun facts helpful diagrams and text features such as a glossary and index The hands on Think Like a Scientist lab activity aligns with Next Generation Science

Standards NGSS The accompanying 5E lesson plan incorporates writing to increase overall comprehension and concept development and features Step by step instructions with before during and after reading strategies Introductory activities to develop academic vocabulary Learning objectives materials lists and answer key Science safety contract for students and parents

Chemical Changes ,2004 Any change that results in the formation of a new kind of matter is a chemical change Energy is always involved in a chemical change sometimes with spectacular results *Physical and Chemical Properties and Changes* Jenny Karpelenia,2007-01-01 Describes the concepts of chemical reactions and the properties of matter Material Changes and Reactions Chris Oxlade,2007-06-06 Introduces the various ways the chemical elements respond to changes in temperature and pressure and how they react with each other describes the role of such changes in everyday life and suggests related activities Chemical Changes Rebecca L. Johnson,2008-06-01 Physical and chemical properties of matter can change In a chemical reaction reactants change into products Energy is either absorbed or released but matter is always conserved Bonds in reactants are broken and atoms are rearranged to form products The pH scale measures the strength of acids and bases found in many substances

Physical and Chemical Changes Michelle L. Szabo,2001

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