

Chemistry I: Periodicity Worksheet

- 1) Arrange the following elements in order of increasing size: Al, Br, C, K, and Na

100

- 2) Arrange the following elements in order of increasing size: Ca, Rb, P, Ge, and Sr.

100

- 3) Circle the atom or ion in each pair that has the larger radius.

- ✓ **Answer:** **Immerse**

- 4) Select the atom or ion in each pair that has the larger radius.

- ch Error As

- 5) Which of the following groups of elements is arranged correctly in order of increasing ionization energy?
(Answer by circling the correct answer.)

- d) $\text{Ne} < \text{C} < \text{Si} < \text{Li}$

- 6) Arrange the following atoms in order of increasing ionization energy: Li, K, C, and N.

1997

- 7) Arrange the following atoms in order of increasing ionization energy: Si, K, As, and Ca.

100

- 8) Compare the elements Na, Mg, O, and P.

- Background:**

Periodicity Worksheet Answer Key

CL Gary

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'CL Gary' text.

Periodicity Worksheet Answer Key:

Periodicity Worksheet Answer Key Book Review: Unveiling the Power of Words

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