

## 5 - Periodicity Activity

20 19 18 17 16 15 14 13 12 11 10 9 8 7 6 5 4 3 2 1

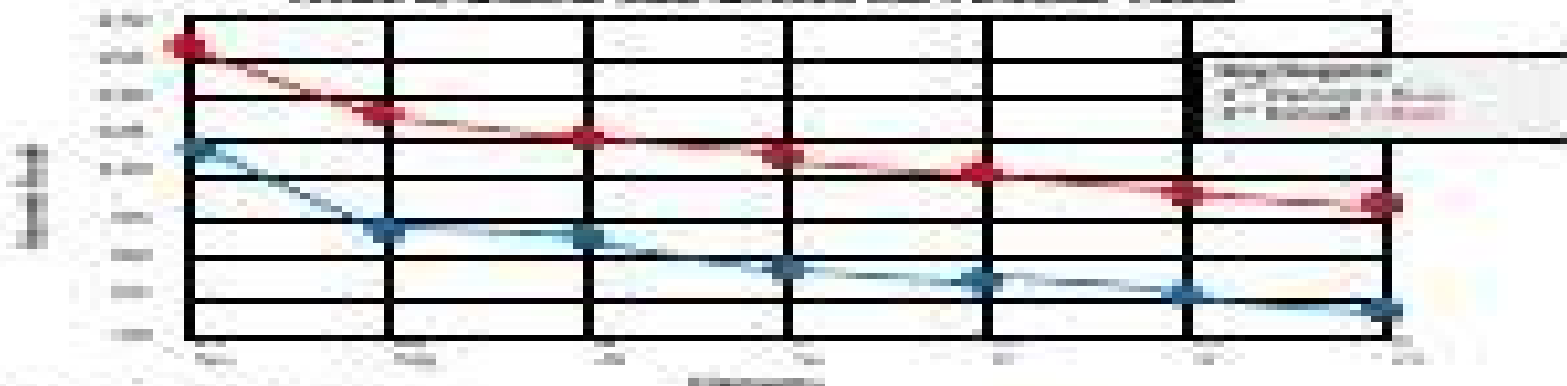
Instructions: Using the graphs above fill the boxes of atomic properties below:

1. Choose which graph best fits the property stated in the graphing sheet of 5 data points (graph above)
2. Is a slight downward trend in this graph Period? If not, use the other graphs
3. Label the box as 'not suitable' if not suitable
4. Answer all questions...

	1	2	3	4	5	6	7	8
1	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
2	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
3	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
4	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
5	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
6	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
7	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
8	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
9	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius
10	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius	Electronegativity	Atomic radius

Now label the trends in the graphing sheet

### Trend in Atomic Size Across the Periodic Table



#### Electronegativity Questions

1. Following the trend in atomic size across the periodic table, decreases
2. This means two atoms that are more electronegative (more electronegative) will have a stronger attraction between the two atoms.

#### Atomic radius Questions

1. Electronegativity increases from left to right, e.g. Na < Cl, Br < F. This is the opposite trend to the trend in atomic size across the periodic table: increases
2. This means two atoms that are more electronegative (more electronegative) will have a stronger attraction between the two atoms.

# Periodicity Activity Okstate Answer Key

**Michael Seilmaier**



**Periodicity Activity Okstate Answer Key:**

The book delves into Periodicity Activity Okstate Answer Key. Periodicity Activity Okstate Answer Key is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Periodicity Activity Okstate Answer Key, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Periodicity Activity Okstate Answer Key
- Chapter 2: Essential Elements of Periodicity Activity Okstate Answer Key
- Chapter 3: Periodicity Activity Okstate Answer Key in Everyday Life
- Chapter 4: Periodicity Activity Okstate Answer Key in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Periodicity Activity Okstate Answer Key. The first chapter will explore what Periodicity Activity Okstate Answer Key is, why Periodicity Activity Okstate Answer Key is vital, and how to effectively learn about Periodicity Activity Okstate Answer Key.

3. In chapter 2, this book will delve into the foundational concepts of Periodicity Activity Okstate Answer Key. The second chapter will elucidate the essential principles that must be understood to grasp Periodicity Activity Okstate Answer Key in its entirety.

4. In chapter 3, the author will examine the practical applications of Periodicity Activity Okstate Answer Key in daily life. This chapter will showcase real-world examples of how Periodicity Activity Okstate Answer Key can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of Periodicity Activity Okstate Answer Key in specific contexts. This chapter will explore how Periodicity Activity Okstate Answer Key is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Periodicity Activity Okstate Answer Key. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Periodicity Activity Okstate Answer Key.

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