

Periodic Trends – Atomic Mass

- The **atomic mass** describes the mass of each individual element.
- **Atomic mass** increases moving **down** each group and **right** across a period.

Instructions: In the periodic table below, draw two arrows representing the direction that atomic mass **increases** across groups and periods.

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Instructions: In each question, identify the element with the **greatest** atomic mass:

- 1) Chlorine (Cl), Iodine (I), Bromine (Br), Fluorine (F) Iodine
- 2) Oxygen (O), Carbon (C), Nitrogen (N), Boron (B) Oxygen
- 3) Lithium (Li), Silicon (Si), Sulfur (Br), Xenon (Xe) Xenon

Instructions: In each question, list the elements from **least** to **greatest** atomic mass:

- 4) Magnesium (Mg), Beryllium (Be), Barium (Ba) Be, Mg, Ba
- 5) Gallium (Ga), Selenium (Se), Potassium (K) K, Ga, Se

Periodic Trends Worksheet Extension Questions

Joann H. Lee



Periodic Trends Worksheet Extension Questions:

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

Online Searching Joann H. Lee, 1989

Describes the management of an online searching service how to set up implement and administer on online reference program Annotation copyrighted by Book News Inc Portland OR

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